

Science (086)
Class X
Sample Question Paper 2022-23

Max. Marks: 80

Time Allowed: 3 hours

General Instructions:

- i. This question paper consists of 39 questions in 5 sections.
- ii. All questions are compulsory. However, an internal choice is provided in some questions. A student is expected to attempt only one of these questions.
- iii. **Section A** consists of 20 objective type questions carrying 1 mark each.
- iv. **Section B** consists of 6 Very Short questions carrying 02 marks each. Answers to these questions should be in the range of 30 to 50 words.
- v. **Section C** consists of 7 Short Answer type questions carrying 03 marks each. Answers to these questions should be in the range of 50 to 80 words.
- vi. **Section D** consists of 3 Long Answer type questions carrying 05 marks each. Answer to these questions should be in the range of 80 to 120 words.
- vii. **Section E** consists of 3 source-based/case-based units of assessment of 04 marks each with sub-parts.

SECTION - A

Select and write one most appropriate option out of the four options given for each of the questions 1 – 20

Q. No	Questions	Marks
1	<p>The change in colour of the moist litmus paper in the given set up is due to</p>  <p>i. presence of acid ii. presence of base iii. presence of $H^+(aq)$ in the solution iv. presence of Litmus which acts as an indicator (a) i and ii (b) Only ii (c) Only iii (d) Only iv.</p>	1
2	<p>In the redox reaction $MnO_2 + 4HCl \rightarrow MnCl_2 + 2H_2O + Cl_2$ (a) MnO_2 is reduced to $MnCl_2$ & HCl is oxidized to H_2O (b) MnO_2 is reduced to $MnCl_2$ & HCl is oxidized to Cl_2</p>	1

Nbuilding Science Paper 23

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NBS Building Science Series, 1974 *Building Science Abstracts*, 1929 *Building Science Series*, 1970-11 **NBS Building Science Series** United States. National Bureau of Standards, 1975 **NIST Serial Holdings** National Institute of Standards and Technology (U.S.), 2002 *Report of the Building Research Board with the Report of the Director of Building Research* Great Britain. Building Research Board, 1940 *Building*, 1928 *Monthly Catalog, United States Public Documents* United States. Superintendent of Documents, 1973 February issue includes Appendix entitled Directory of United States Government periodicals and subscription publications September issue includes List of depository libraries June and December issues include semiannual index *Environmental and Human Impact of Buildings* Ligia Moga, Teodora M. Şoimoşan, 2021-02-24 Featuring research on topics such as low energy buildings concepts construction materials and technology hybrid energy systems energy balance and wellbeing this book meets the expectations of academicians specialists and researchers in the field along with the scholars seeking coverage on buildings environmental and human impact It presents an integrated approach to the buildings energetic aspects from the perspective of environmental impact together with the indoor wellbeing In this respect the chapters include state of the art case studies as well as research results that validate the raised hypotheses The book integrates topics related to buildings performance approached by researchers with different backgrounds within the civil engineering domain i.e. achieved energetics performances obstacles restrictions and limitations issues within design and optimization processes including the new perspectives in the buildings energy sector

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Urban Heat Island and its Impact on Building Energy Consumption Green Roofs in Buildings Thermal and Environmental
Behaviour Building Earth Contact Heat Transfer *A Genealogy of Tropical Architecture* Jiat-Hwee Chang, 2016-04-28 A
Genealogy of Tropical Architecture traces the origins of tropical architecture to nineteenth century British colonial
architectural knowledge and practices It uncovers how systematic knowledge and practices on building and environmental
technologies in the tropics were linked to military technologies medical theories and sanitary practices and were manifested
in colonial building types such as military barracks hospitals and housing It also explores the various ways these colonial
knowledge and practices shaped post war techno scientific research and education in climatic design and modern tropical
architecture Drawing on the interdisciplinary scholarships on postcolonial studies science studies and environmental history
Jiat Hwee Chang argues that tropical architecture was inextricably entangled with the socio cultural constructions of tropical
nature and the politics of colonial governance and postcolonial development in the British colonial and post colonial networks
By bringing to light new historical materials through formidable research and tracing the history of tropical architecture
beyond what is widely considered today as its founding moment in the mid twentieth century this important and original book
revises our understanding of colonial built environment It also provides a new historical framework that significantly bears
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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the iron metal dissolves

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web purpose to enhance the understanding of stoichiometry a reaction between iron and copper ii sulfate solution will be conducted this will help you to differentiate limiting and excess reactant in a chemical reaction finally the theoretical and

percent yield of this reaction will be calculated

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web purpose in this experiment you will observe a single replacement reaction you will also use stoichiometry to predict what the theoretical yield of product is and calculate a percent yield reaction iron copper ii sulfate copper iron ii sulfate pre lab questions write the balanced equation

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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate CuSO_4 and 2 0153 grams of iron powder Fe the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web stoichiometry lab the reaction of iron with copper ii sulfate the study of stoichiometry deals with the calculation of quantities in a chemical reaction how much product will be produced how much reactant do you need to make that much product

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web description great lab for use during stoichiometry unit in chemistry reinforces ideas of mole ratios and stoichiometry calculations by measuring the mass of iron that reacts and the mass of copper metal produced students can calculate the ratio of moles of reactant to moles of product this experimental mole ratio from lab can be compared

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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stirring the solution adding water then removing water adding acetone heating the acetone on low

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