

Chemistry Honors Final Study Guide

- Semester 2 -

Solids, Liquids, and Solutions | Chapters 14 & 15:

⇒ Intermolecular Forces

- **Intermolecular Forces:** forces that hold molecules together in liquids and solids (are weaker than intramolecular forces)
 - Types of intramolecular forces are ionic bonds, covalent bonds, and metallic bonds.
 - The 3 intermolecular forces are *dipole-dipole forces*, *dispersion forces*, and *hydrogen bonding*.

✧ Dipole-Dipole Forces

- **Dipole-Dipole Forces:** attractions between opposite charges of neighboring (permanent) dipoles.
- Polar molecules are attracted to opposite ends of each other through dipole-dipole forces
- This intermolecular force is only possible with polar molecules

✧ Ion-Dipole Forces

- **Ion Dipole Force:** an attractive force that results from the electrostatic attraction between an ion and a neutral molecule that has a dipole

✧ Dispersion Forces

- **Induced Dipole:** a dipole created by the presence of a neighboring dipole
- **Dispersion Forces:** the force of attraction between induced dipoles (occur whenever electrons are involved)
- **Polarizability:** the ease with which the electron cloud of an atom or molecule is distorted
 - The respective trend is atomic (or molar) mass/radius/weight and it is the exact opposite of the ΔEN trend on the periodic table

Semester 2 Chemistry Honors Review Guide

G Psacharopoulos



Semester 2 Chemistry Honors Review Guide:

The Enigmatic Realm of **Semester 2 Chemistry Honors Review Guide**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Semester 2 Chemistry Honors Review Guide** a literary masterpiece penned by a renowned author, readers embark on a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of people who partake in its reading experience.

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