

Separation of a Dye Mixture Using Chromatography

Lab Experiment Part 1

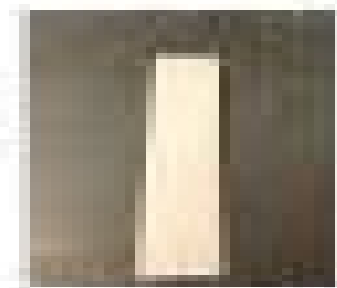
Objectives

1. Understand the principles of chromatography.
2. Perform a separation of a dye mixture using chromatography.
3. Calculate the retention time (R_f) of the components.
4. Identify the components of the mixture.
5. Understand the factors affecting the separation.
6. Calculate the partition coefficient (K_d).
7. Understand the applications of chromatography.
8. Calculate the resolution (R_s).
9. Understand the importance of chromatography in analytical chemistry.
10. Calculate the peak area.
11. Understand the factors affecting the peak area.
12. Calculate the peak height.
13. Understand the factors affecting the peak height.
14. Calculate the peak width.
15. Understand the factors affecting the peak width.

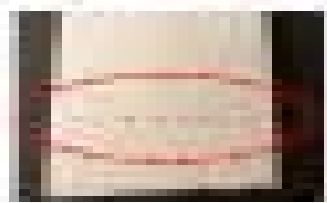
Introduction

Chromatography is a technique used to separate the components of a mixture. It is based on the principle of differential partitioning between a stationary phase and a mobile phase.

1. The stationary phase is a solid or a liquid supported on a solid support.
2. The mobile phase is a liquid or a gas that moves through the stationary phase.



3. The components of the mixture are separated based on their partition coefficient (K_d).
4. The retention time (R_f) is the ratio of the distance traveled by the component to the distance traveled by the solvent front.
5. The resolution (R_s) is the ratio of the distance between the peaks to the width of the peaks.
6. The peak area is the area under the peak.
7. The peak height is the maximum height of the peak.
8. The peak width is the width of the peak at the base.



Separation Of A Dye Mixture Using Chromatography Lab Answers

RD Boyd



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