

Separation of a Dye Mixture Using Chromatography

Lab Experiment Part 1

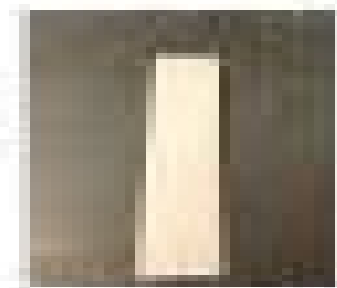
Objectives

1. Understand the principles of chromatography.
2. Perform a separation of a dye mixture using chromatography.
3. Calculate the retention time (R_f) of the components.
4. Identify the components of the mixture.
5. Understand the importance of chromatography in analytical chemistry.
6. Develop a systematic approach to solving problems.
7. Understand the role of chromatography in the pharmaceutical industry.
8. Understand the role of chromatography in environmental science.
9. Understand the role of chromatography in forensic science.
10. Understand the role of chromatography in food science.
11. Understand the role of chromatography in clinical chemistry.
12. Understand the role of chromatography in biotechnology.
13. Understand the role of chromatography in materials science.
14. Understand the role of chromatography in agriculture.
15. Understand the role of chromatography in the chemical industry.

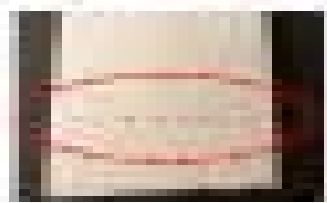
Introduction

Chromatography is a technique used to separate mixtures into their individual components. It is based on the principle of differential partitioning between a stationary phase and a mobile phase.

1. The stationary phase is a solid or a liquid supported on a solid support.
2. The mobile phase is a liquid or a gas that moves through the stationary phase.



3. The components of the mixture are separated based on their affinity for the stationary phase.
4. The retention time (R_f) is the distance traveled by the component divided by the distance traveled by the solvent front.
5. The R_f value is a characteristic property of a component and can be used to identify it.



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