



Name: _____
Period: _____

KEY

Nuclear Radiation Worksheet

1. a. What determines whether an atom is stable or unstable?

Its overall size and a difference between proton # and neutron #

- b. How does an unstable atom gain stability?

It undergoes radioactive decay

- c. What part of an atom undergoes change during radioactive decay?

The nucleus!

2. Which of the human senses can detect radiation?

Our eyes can detect colors, but we have no senses to detect alpha, beta, or gamma radiation

3. Explain the difference between an isotope and a radioactive isotope.

An isotope has a different # of protons & neutrons. Radioactive is larger, unstable, and has a different # of p

4. Distinguish between alpha, beta, and gamma radiation on the basis of the following:

Particle	Mass	Charge	Penetrating power
Alpha	4 amu, 2 neutrons, 2 protons	Positive (+)	Stopped by paper or paper
Beta	0 Neutrons (electron)	Negative (-)	Stopped by sheets of tin foil or thick magazine
Gamma	0 Neutrons	Neutral	Stopped by several inches of lead or feet of concrete

5. How are the mass number and atomic number of a nucleus affected by the loss of the following?

Particle	Mass Number Change	Atomic Number Change
Alpha	-4	-2
Beta	No change	+1
Gamma	No change	No change

6. Complete the equations for the decay processes below by supplying the missing atomic numbers and mass numbers.



7. Complete the following nuclear equations:



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Liyong Dong



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Table of Contents Snyder 2007 Nuclear Radiation Worksheet Answer Key

1. Understanding the eBook Snyder 2007 Nuclear Radiation Worksheet Answer Key
 - The Rise of Digital Reading Snyder 2007 Nuclear Radiation Worksheet Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Snyder 2007 Nuclear Radiation Worksheet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Snyder 2007 Nuclear Radiation Worksheet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Snyder 2007 Nuclear Radiation Worksheet Answer Key
 - Personalized Recommendations
 - Snyder 2007 Nuclear Radiation Worksheet Answer Key User Reviews and Ratings
 - Snyder 2007 Nuclear Radiation Worksheet Answer Key and Bestseller Lists
5. Accessing Snyder 2007 Nuclear Radiation Worksheet Answer Key Free and Paid eBooks
 - Snyder 2007 Nuclear Radiation Worksheet Answer Key Public Domain eBooks
 - Snyder 2007 Nuclear Radiation Worksheet Answer Key eBook Subscription Services

- Snyder 2007 Nuclear Radiation Worksheet Answer Key Budget-Friendly Options
- 6. Navigating Snyder 2007 Nuclear Radiation Worksheet Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Snyder 2007 Nuclear Radiation Worksheet Answer Key Compatibility with Devices
 - Snyder 2007 Nuclear Radiation Worksheet Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Snyder 2007 Nuclear Radiation Worksheet Answer Key
 - Highlighting and Note-Taking Snyder 2007 Nuclear Radiation Worksheet Answer Key
 - Interactive Elements Snyder 2007 Nuclear Radiation Worksheet Answer Key
- 8. Staying Engaged with Snyder 2007 Nuclear Radiation Worksheet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Snyder 2007 Nuclear Radiation Worksheet Answer Key
- 9. Balancing eBooks and Physical Books Snyder 2007 Nuclear Radiation Worksheet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Snyder 2007 Nuclear Radiation Worksheet Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Snyder 2007 Nuclear Radiation Worksheet Answer Key
 - Setting Reading Goals Snyder 2007 Nuclear Radiation Worksheet Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Snyder 2007 Nuclear Radiation Worksheet Answer Key
 - Fact-Checking eBook Content of Snyder 2007 Nuclear Radiation Worksheet Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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