

Separation of a Dye Mixture Using Chromatography

Pre-Lab Assignment

- Figure 1 is a sample paper chromatogram for three samples: A, B and C. Label the drawing with the following items: the stationary phase, the mobile phase and the solvent front.

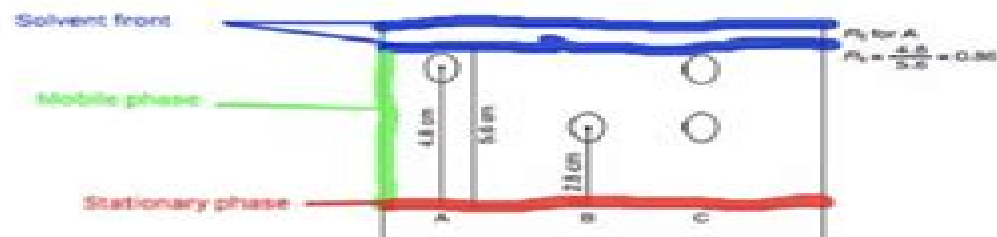


Figure 1.

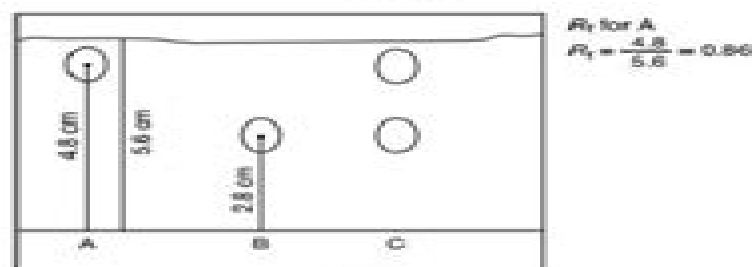


Figure 1.

- Calculate the Rf value for the spot in sample B using sample A as an example. $R_f \text{ B} = .5$
- Sample C gave two spots on the paper chromatogram. What does this tell you about the composition of the sample? *It means that there is two different compounds in it.*
- Based on the Rf values of samples A and B, what can you conclude about the intermolecular attractions both samples have for the eluent and the paper? *Sample A is less polar and sample B is more polar.*

Background

The entire palette of artificial food colors is derived from just seven dyes certified by the FDA for use in food, drugs, and cosmetics. How can these FD&C dyes be identified in a mixture? How do the molecular structures of the dye molecules influence their properties, including their relative solubility or affinity for different solvents?

The use of color additives increased dramatically in the United States in the second half of the nineteenth century. As the economy became more industrial, demographics shifted, fewer

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Answers Flinn

Hannah Lynn Barks



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Journal of Applied Chemistry, 1970 Separation and Identification of Complex Mixtures Using Chromatography Mass Spectrometry Hannah Lynn Barks, 2010 Here for the first time the formation of adenine hypoxanthine and guanine from formamide solutions with heating only to 130 degrees C and UV irradiation in the absence of minerals or inorganic salts is shown using LC MS MS as the analysis technique The thesis goes on to demonstrate that the product distributions change drastically when the temperature is increased to 160 degrees C from 130 degrees C specifically that the amount of hypoxanthine increases with the addition of UV light and the amount of adenine increases with an increase in temperature Along with showing the formation of purines in these reactions the identification of pyrimidines was also achieved by GCxGC MS GCxGC MS was also used to study additional samples specifically bio oils where the type of compounds in the samples were easily identifiable which allowed for a direct comparison between different types of bio oils e g Douglas fir bark Southern pine bark and a Southern pine bark wood mixture

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