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web correct answer option b q3 the image shows some chemical reactions which option identifies the reactants and products of the reactions correct answer option c q4 a student performs an experiment to form aluminium chloride from aluminium and chlorine

[chemical reactions and equations class 10 extra questions with answers](#) - Jul 09 2022

web apr 19 2023 chemical reactions and equations extra questions very short answer type question 1 why is the combustion of liquified petroleum gas lpg a chemical change answer combustion of liquified petroleum gas lpg is a chemical change because after its combustion a new substance is formed and cannot be turned back into

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web q5 identify the reducing agent in the following reactions a $4\text{NH}_3 + 5\text{O}_2 \rightarrow 4\text{NO} + 6\text{H}_2\text{O}$ b $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ c $\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$ d $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ answer a here ammonia NH_3 is the reducing agent b here water H_2O is the reducing agent