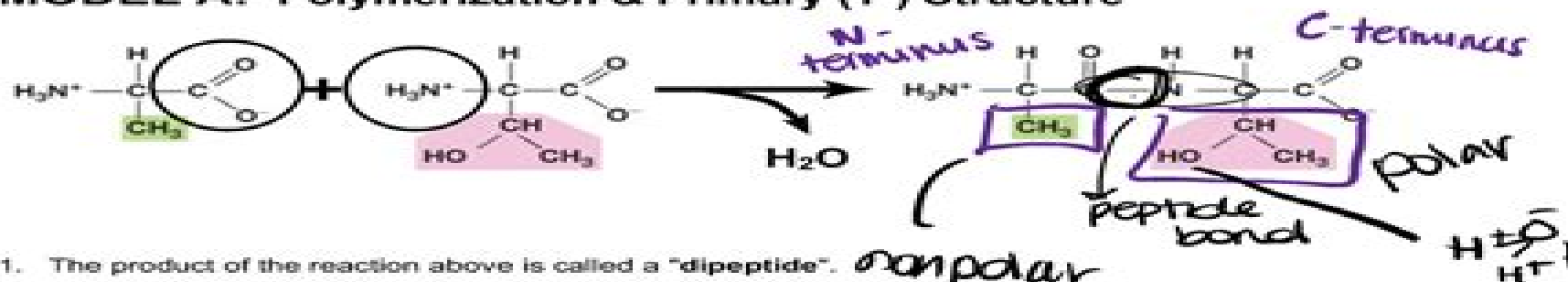
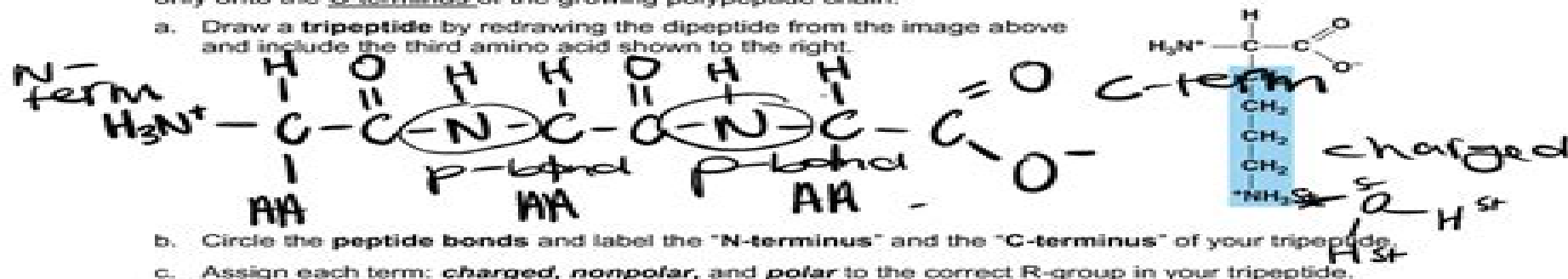


MODEL A: Polymerization & Primary (1°) Structure



1. The product of the reaction above is called a "dipeptide".
 - a. Which two functional groups are involved in this reaction? Circle and label these two groups on the reactants. **Carboxyl and amino group**
 - b. Circle and label the new **peptide bond** formed between the two amino acids.
 - c. Does a peptide bond involve the amino acid "**backbone**" or **R-groups**? (Circle one.)
 - d. Draw a square around each **amino acid** in the dipeptide. An amino acid that is part of a polypeptide chain is referred to as a "**residue**."
 - e. Label the "**N-terminus**" and the "**C-terminus**" (also known as the amino- and carboxy-termini) of the dipeptide molecule. ("terminus"="end")
2. The addition (**polymerization**) of amino acid **monomers** is directional, with new amino acids added only onto the C-terminus of the growing polypeptide chain.



- a. Draw a **tripeptide** by redrawing the dipeptide from the image above and include the third amino acid shown to the right.
- b. Circle the **peptide bonds** and label the "**N-terminus**" and the "**C-terminus**" of your tripeptide.
- c. Assign each term: **charged**, **nonpolar**, and **polar** to the correct R-group in your tripeptide.
- d. Two of your R-groups are **hydrophilic** and can form **hydrogen bonds** with water. Draw two H₂O molecules interacting with your tripeptide, using dotted lines to represent the hydrogen bonds. Label the partial charges of the atoms involved in each hydrogen bond you draw.

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SJ Ball



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