

Limiting Reactant vs Excess Reactant

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DEFINITION

Limiting Reactant

Limiting reactant is the reactant that can limit the formation of the final product in a chemical reaction

Excess Reactant

Excess reactant is the reactant that is present in excess in a reaction mixture

DESCRIPTION

Completely consumed during the chemical reaction

Not all reactant molecules are consumed in the reaction

END OF THE REACTION

No limiting reactant present

Excess reactant is present

EFFECT ON FINAL PRODUCT

Limits the amount of final product formed

No effect on the final product

Pogil Ap Chem Limiting And Excess Reactions

Lifelique

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Limiting Reactant and Percent Yield, 2019 This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant calculate the theoretical yield of a reaction and calculating the percent yield of a reaction

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