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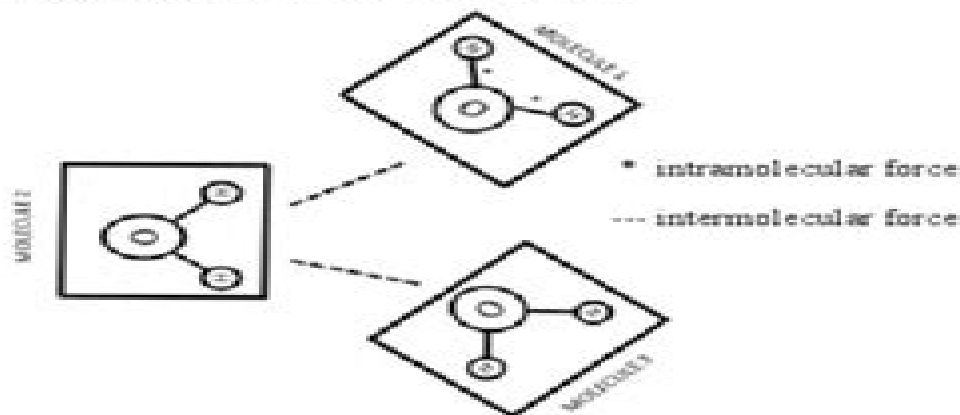
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POGIL: Intermolecular Forces

Model 1: What is an intermolecular force?

As you have learned, matter is made up of discrete particles called **atoms**, which chemically combine to form **molecules**. Molecules do not exist as independent units: in fact, groups of molecules "stick together" in order to form liquids and solids. The forces that hold groups of molecules together are **intermolecular forces**. Without intermolecular forces, the world as we know it would not be the same.

Figure 1: Intramolecular and Intermolecular Forces



Critical Thinking Questions:

1. What specific molecule is represented inside each box in Figure 1?
2. In relation to the box for molecule 1, where do the intramolecular forces exist in Figure 1 – inside the boxes or outside the boxes?
3. Based on the intramolecular forces for molecule 1, draw similar asterisks (*) for the intramolecular forces on the diagram for molecules 2 and 3.
4. In relation to the molecule, where do intramolecular forces tend to occur – within the molecule or outside of the molecule?
5. Two intermolecular forces exist in Figure 1. Where are they positioned relative to the molecules – within the molecules or between the molecules?
6. State the difference between intermolecular and intramolecular forces in terms of where they occur on the molecular level.



Pogil Intermolecular Forces Answer Key

NARAYAN CHANGDER



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Table of Contents Pogil Intermolecular Forces Answer Key

1. Understanding the eBook Pogil Intermolecular Forces Answer Key
 - The Rise of Digital Reading Pogil Intermolecular Forces Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Pogil Intermolecular Forces Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pogil Intermolecular Forces Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pogil Intermolecular Forces Answer Key
 - Personalized Recommendations
 - Pogil Intermolecular Forces Answer Key User Reviews and Ratings
 - Pogil Intermolecular Forces Answer Key and Bestseller Lists
5. Accessing Pogil Intermolecular Forces Answer Key Free and Paid eBooks

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 - Pogil Intermolecular Forces Answer Key Budget-Friendly Options
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 - Pogil Intermolecular Forces Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
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 - Highlighting and Note-Taking Pogil Intermolecular Forces Answer Key
 - Interactive Elements Pogil Intermolecular Forces Answer Key
8. Staying Engaged with Pogil Intermolecular Forces Answer Key
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pogil Intermolecular Forces Answer Key
9. Balancing eBooks and Physical Books Pogil Intermolecular Forces Answer Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pogil Intermolecular Forces Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pogil Intermolecular Forces Answer Key
- Setting Reading Goals Pogil Intermolecular Forces Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pogil Intermolecular Forces Answer Key
- Fact-Checking eBook Content of Pogil Intermolecular Forces Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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