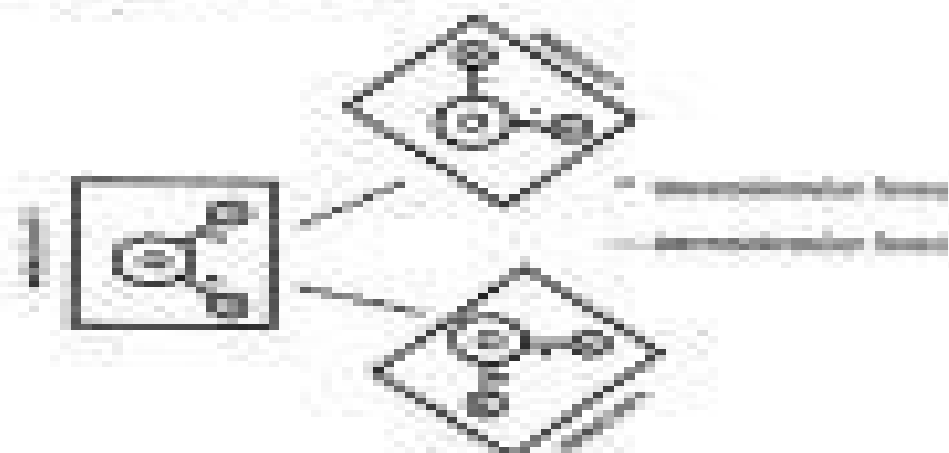


# POGIL: Intermolecular Forces

## Model 1: What is an Intermolecular Force?

As you have learned, matter is made up of discrete particles called atoms which chemically combine to form molecules. Molecules do not exist as independent units; in fact, groups of molecules "stick together" in order to form liquids and solids. The forces that hold groups of molecules together are Intermolecular Forces. Without intermolecular forces, the world as we know it would not be the same.

Figure 1: Intermolecular and Intramolecular Forces



## Critical Thinking Questions

1. What specific molecule is represented inside each box in Figure 1?  
 $\text{H}_2\text{O}$
2. In relation to the box for molecule 1, where do the intermolecular forces exist in Figure 1—inside the boxes or outside the boxes?  
 $\text{outside the boxes}$
3. Based on the intermolecular forces for molecule 1, draw similar patterns (?) for the intermolecular forces of the (diagram the molecules 2 and 3).
4. In relation to the molecule, where do intramolecular forces tend to occur—within the molecule or outside of the molecule?  
 $\text{within the molecule}$
5. Draw intermolecular forces under in Figure 1. Where are they positioned relative to the molecules—within the molecules or between the molecules?  
 $\text{between the molecules}$

6. State the difference between intermolecular and intramolecular forces in terms of where they exist on the molecular level.

Intermolecular = happens  $\text{between}$  separate molecules, like boiling water vs. freezing water  
 Intramolecular = happens  $\text{within}$  each molecule, like tearing apart a molecule  
 (happening on the inside-out)

# Pogil Intermolecular Forces Key

**Yijin Wang**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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