

Oxidation and Reduction

When copper(II) nitrate is transformed in a chemical reaction?

Q1a)?

Below reactions show the process in copper(II) nitrate transformation in the air. Copper gets oxidized in atmospheric gases present in the air. Metals react or corrode in the presence of air and water. Atmospheric gases responsible for corrosion are dissolved in water and react with the metals to produce more harmful substances. All of these reactions are examples of oxidation and reduction, collectively known as redox reactions. In this activity, you will explore what is happening in the above listed redox reactions.

Model 1 – Redox Reactions

Redox Reactions:



Redox Reactions:



1. What are typical reactions (reactions in Model 1)?

Redox reactions and reduction reactions.

2. Explain the redox and reduction reactions in Model 1. What does oxidation/reduction in the redox reactions (the redox) allow you to identify about the redox reactions? It gives you specific examples from Model 1 as support your answer.

There is no obvious distinguishing feature between oxidation and reduction reactions.

Pogil Lab Oxidation Answer Key

RS Peters



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Analytical Chemistry Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the BerriLambert atomic and molecular absorption processes vibrational modes mass spectra interpretation and much more **Oxidation and Reduction** ,1992 Oxidation Numbers of Fractionally Charged Atoms Los Alamos National Laboratory, G. Zweig, K.S. Lackner, 1983 **The Oxidation of Aqueous Potassium Sulfite Solution** Kuan-in Shiau, 1975 **Keys to Oxidation-reduction** ,1974

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