

POGIL: When an Atom Drinks Red Bull

Honors Chemistry
Unit 07 - The Atom
Page 1 of 3

Name: _____

Date: _____

Model 1: White Light



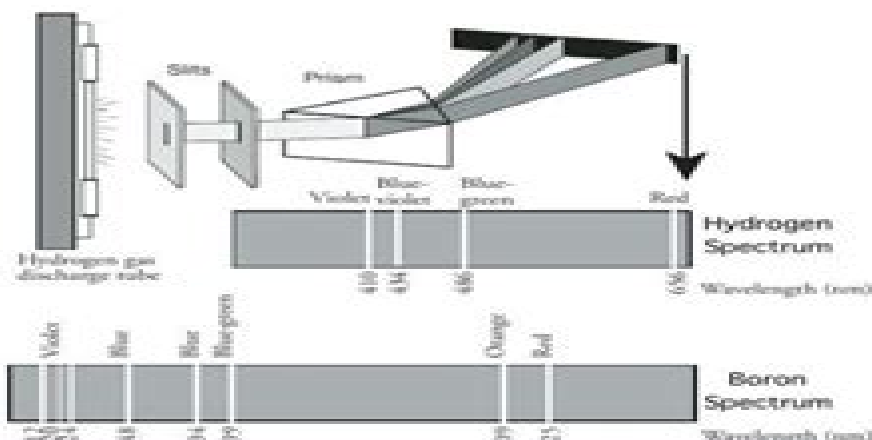
Prism

Color	Photon Energy ($\times 10^{-19}$) (J)	Wavelength Range (nm)	Speed (m/s)
Reds	269–318	625–740	3.00×10^8
Oranges	318–337	590–625	3.00×10^8
Yellows	337–352	565–590	3.00×10^8
Greens	352–382	520–565	3.00×10^8
Blues	382–452	440–520	3.00×10^8
Violens	452–523	380–440	3.00×10^8

- Do all colors of light travel at the same speed? If no, which color has the greatest speed?
- Do all colors of light have the same energy? If no, which color has the highest energy and which color has the lowest?
- Consider the light illustrated in Model 1.
 - Which color corresponds to the longest wavelength?
 - Which color corresponds to the shortest wavelength?
 - Write a sentence that describes the relationship between wavelength and energy of light.

Model 2: Emission Spectra for Hydrogen and Boron Atoms

- Consider the boron spectrum in Model 2. Thinking about your answers to question 3, ...
 - which color line of light corresponds to the shortest wavelength?
 - which color line of light corresponds to the longest wavelength?
- Consider the hydrogen spectrum in Model 2. Thinking about your answers to question 2, ...
 - which color line of light has the most energy?
 - which color line of light has the least energy?



Pogil Project Flinn Science Electron Energy And Light

ML Morrison



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