

Balancing Chemical Equations

NAME _____

HOUR _____

STANDARD 1 LEARNING MODULE 10

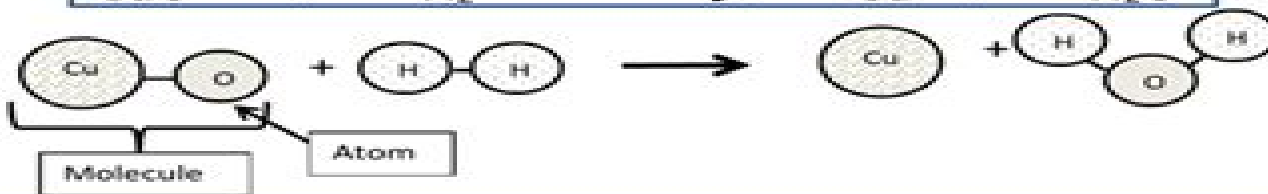
Background Information and Why: Chemical reactions can be described by chemical equations. Equations means equal. One each side of the equal side, also known as "yields", the number AND type of atoms are exactly the same. The Law of Conservation of Mass states that the atoms cannot be created or destroyed. The atoms can only be rearranged into different molecules/substances. Atoms are the building blocks of molecules. Therefore, in order to obey The Law of Conservation of Mass, the number of atoms for each element must be the same in the reactants (left side) as it is in the products (right side). In order to balance equations we cannot split or add to existing molecules. We can change the coefficient, adding molecules. We cannot change the subscripts, the small numbers, because they determine how many atoms are in a molecule.

REACTION A

Chemical Equation



Atomic Representation



REACTION B

Chemical Equation



Atomic Representation



Questions

0.) Label one more molecule and one more atom in EACH reaction. Use the ones done above to help you. Pick another atom draw an arrow to it. Label it "atom". Then circle or "bracket" another molecule and label it "molecule".

1.) How many of each type of atom are represented in each side of reaction A? Count them up and fill out the table below.

# Atoms in Reactants	Atoms in Products
Cu- ONE	Cu- ONE
O- ONE	O- ONE
H- TWO	H- TWO

Pogil Types Of Chemical Reactions Balancing Equations

Jake Mariska



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Edmondson presents basic information for high school students on chemical equations and reactions Teachers Net provides the information as part of the Teachers Net Lesson Exchange online resource **Real World Examples of Four Types of Chemical Reactions** , James W Smith developed Real World Examples of Four Types of Chemical Reactions a physical science and chemistry lesson for 9th grade students The students use the Internet to locate information about industrial pharmaceutical and environmental operations that use examples of the four general classes of chemical reactions These reactions are synthesis decomposition single replacement reactions and double replacement reactions The Louisiana Challenge Grant provides the lesson online as part of the Louisiana Challenge collection of activities for the K 12 classroom

Writing and Balancing Chemical Equations ,2016 What happens when you combine two or more elements Through a variety of practice problems you learn to identify when a chemical reaction has occurred how to write chemical equations and how to balance equations to conserve the atoms **Chemical Equations** Lifelique,2019 This lesson plan covers describing chemical reactions using word equations knowing the correct symbols to use to write skeleton equations for chemical reactions and balancing chemical equations *It's in the Equation! Law of Conservation of Mass and Chemical Equations | Grade 6-8 Physical Science* Baby Professor,2024-04-15 Discover the foundational principles of chemical reactions with this insightful book perfect for middle school educators homeschooling parents and school librarians Explore the Law of Conservation of Mass and Learn about the intricacies of chemical equations highlighting their critical role in the US STEM curriculum From understanding mass conservation to mastering the art of balancing equations this book equips young learners with the essential knowledge to navigate the world of chemistry Dive into this must read for a comprehensive grasp of how substances transform in chemical reactions

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