

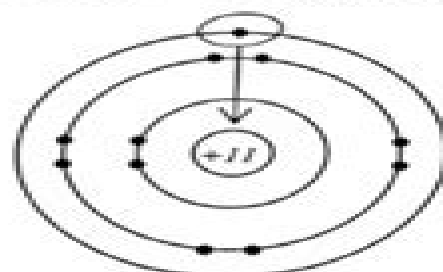
## Advanced Periodic Trends

How do the particles in an atom interact to dictate periodic properties of elements?

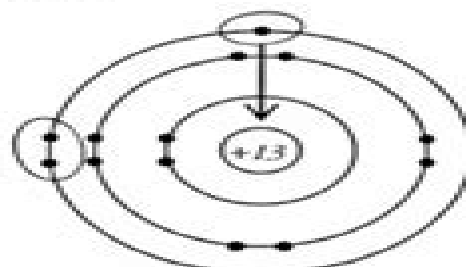
### Why?

Previously you have learned about Coulombic attraction and how it governs the trends in properties among the elements of the Periodic Table. For example, as you move across a period, atoms have more protons in the nucleus, which pulls the electrons in tighter making smaller, more electronegative atoms. As you proceed down a column of the periodic table, the valence electrons get further away, and the atoms get larger and less electronegative. Inquisitive students will always ask, "If more protons in the nucleus pull electrons in tighter, then why don't atoms get smaller when going down a column?" That is a good question, and one that you will investigate in this activity.

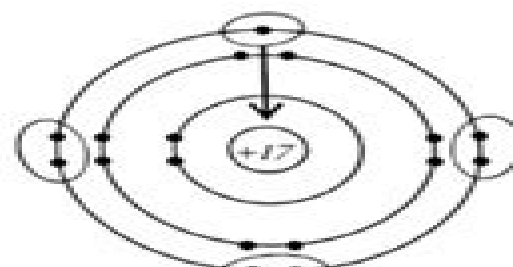
### Model 1 – Period 3 Elements



Sodium



Aluminum



Chlorine

Atomic Number	11	13	17
Number of Electrons	11	13	17
Number of Valence Electrons	1	3	7
Core Charge	+1	+3	+7
Atomic Radius	186 pm	143 pm	99 pm
1st Ionization Energy	496 kJ/mol	578 kJ/mol	1251 kJ/mol
Electronegativity	0.9	1.5	3.0

1. Describe the relationship of the three elements in Model 1 with regard to their relative positions on the periodic table.

*The three elements are in the same row of the periodic table.*

2. Refer to a periodic table to complete the first three rows of the table in Model 1.

*See Model 1.*

# Pogil Chemistry Activities Periodic Trends Activities

**Flinn Scientific**



## **Pogil Chemistry Activities Periodic Trends Activities:**

Chemistry Education Javier García-Martínez, Elena Serrano-Torregrosa, 2015-05-04 Winner of the CHOICE Outstanding Academic Title 2017 Award This comprehensive collection of top level contributions provides a thorough review of the vibrant field of chemistry education Highly experienced chemistry professors and education experts cover the latest developments in chemistry learning and teaching as well as the pivotal role of chemistry for shaping a more sustainable future Adopting a practice oriented approach the current challenges and opportunities posed by chemistry education are critically discussed highlighting the pitfalls that can occur in teaching chemistry and how to circumvent them The main topics discussed include best practices project based education blended learning and the role of technology including e learning and science visualization Hands on recommendations on how to optimally implement innovative strategies of teaching chemistry at university and high school levels make this book an essential resource for anybody interested in either teaching or learning chemistry more effectively from experience chemistry professors to secondary school teachers from educators with no formal training in didactics to frustrated chemistry students

**POGIL Activities for AP\* Chemistry** Flinn Scientific, 2014

**Selected Activities from Foundations of Chemistry** David Hanson, 2017-07-25

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**Chemistry: Molecules, Matter, and Change Media Activities Book** Loretta Jones, Carl Hoeger, Peter William Atkins, Regina Schoenfield-Tacher, 2000-01-15 The Media Activity Book MAB for Jones Atkins Chemistry molecules matter and change contains chapters with lists and descriptions of some of the media available as you study the chapter Each activity begin with a specific textbook reference Then you are given a time estimate of how long it will take to use the media An M media icon in the margin of the textbook means that media exists to support that area of text The media is found in three different places on the website and on two CDs

*Atoms, Molecules & Elements: What Are Compounds? Gr. 5-8* George Graybill, 2015-10-01 This is the chapter slice What Are Compounds from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives

**Simply Science**, 1997

**Explorations in Conceptual Chemistry** Jeffrey Paradis, 2006-04 Designed to give readers a conceptual appreciation for chemistry this manual offers a set of hands on activities on topics such as measurement temperature

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**Integrated Physics and Chemistry, Chapter 1, Activities** Paradigm Accelerated Curriculum,2005-01-01 Key topics Periodic Table of the Elements money metals nonmetals compounds formulas atomic weights heat measuring temperatures Robert Boyle Democritus Lavoisier Proust Dalton Rumford IPC consists of twelve chapters of text and twelve companion student activity books This course introduces students to the people places and principles of physics and chemistry It is written by internationally respected scientist author John Hudson Tiner who applies the vignette approach which effectively draws readers into the text and holds attention The author and editors have deliberately avoided complex mathematical equations in order to entice students into high school level science Focus is on the people who contributed to development of the Periodic Table of the Elements Students learn to read and apply the Table while gaining insight into basic chemistry and physics This is one of our most popular courses among high school students especially those who have a history of under

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Learning about Chemistry Concepts. Superific Science Book VIII. A Good Apple Science Activity Book for Grades 5-8 Lorraine Conway,1983 Provides instructions for a series of experiments that illustrate the principles of chemistry including the nature of matter atoms and molecules chemical changes and the elements and how they interact

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## **Table of Contents Pogil Chemistry Activities Periodic Trends Activities**

1. Understanding the eBook Pogil Chemistry Activities Periodic Trends Activities
  - The Rise of Digital Reading Pogil Chemistry Activities Periodic Trends Activities
  - Advantages of eBooks Over Traditional Books
2. Identifying Pogil Chemistry Activities Periodic Trends Activities
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Pogil Chemistry Activities Periodic Trends Activities
  - User-Friendly Interface
4. Exploring eBook Recommendations from Pogil Chemistry Activities Periodic Trends Activities
  - Personalized Recommendations
  - Pogil Chemistry Activities Periodic Trends Activities User Reviews and Ratings
  - Pogil Chemistry Activities Periodic Trends Activities and Bestseller Lists
5. Accessing Pogil Chemistry Activities Periodic Trends Activities Free and Paid eBooks
  - Pogil Chemistry Activities Periodic Trends Activities Public Domain eBooks
  - Pogil Chemistry Activities Periodic Trends Activities eBook Subscription Services

- Pogil Chemistry Activities Periodic Trends Activities Budget-Friendly Options
- 6. Navigating Pogil Chemistry Activities Periodic Trends Activities eBook Formats
  - ePub, PDF, MOBI, and More
  - Pogil Chemistry Activities Periodic Trends Activities Compatibility with Devices
  - Pogil Chemistry Activities Periodic Trends Activities Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Pogil Chemistry Activities Periodic Trends Activities
  - Highlighting and Note-Taking Pogil Chemistry Activities Periodic Trends Activities
  - Interactive Elements Pogil Chemistry Activities Periodic Trends Activities
- 8. Staying Engaged with Pogil Chemistry Activities Periodic Trends Activities
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Pogil Chemistry Activities Periodic Trends Activities
- 9. Balancing eBooks and Physical Books Pogil Chemistry Activities Periodic Trends Activities
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Pogil Chemistry Activities Periodic Trends Activities
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Pogil Chemistry Activities Periodic Trends Activities
  - Setting Reading Goals Pogil Chemistry Activities Periodic Trends Activities
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pogil Chemistry Activities Periodic Trends Activities
  - Fact-Checking eBook Content of Pogil Chemistry Activities Periodic Trends Activities
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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