

Oxidation and Reduction

When copper(II) nitrate is transformed in a chemical reaction?

Q1a)?

Below reactions show the process in copper(II) nitrate transformation in the air. Copper gets oxidized in chemical gases present in the air. Metals react or combine in the presence of air and water. Nitrate(II) ions decompose to give the reaction for decomposition with the use of electron as positive ions. Metals and compounds, PSE of these reactions are examples of oxidation and reduction, reduction happens in metal reactions. In this reaction, you will replace what is happening in the reaction level in metal reactions.

Model 1 – Redox Reactions

Redox Reactions:



Redox Reactions:



1. What are typical reactions in Model 1?

Redox reactions and reduction reactions.

2. Explain the redox and reduction reactions in Model 1. What does the transformation in the redox reactions show about the reaction process? Explain the reaction process in Model 1 as a typical redox reaction.

There is no reaction showing the reaction between the redox reactions and the reaction.

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**Irish Science Teachers' Association.
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