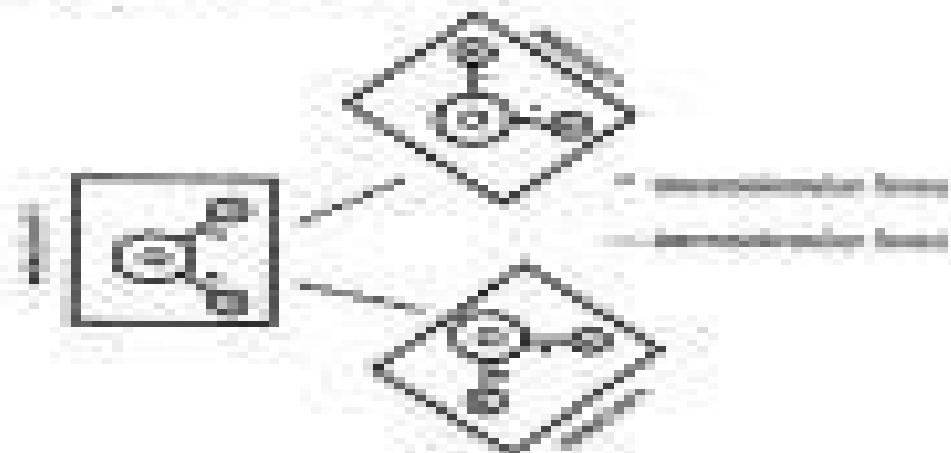


POTENTIAL INTERMOLECULAR FORCES

Model 1: What It Does: Intermolecular Forces

The term "intermolecular" refers to results of all interactions between objects which do not involve the same molecule. Molecules do not want to interact with one other; groups of molecules "want together" in order to form liquids and solids. The forces that hold groups of molecules together are intermolecular forces. Without intermolecular forces, the world we see around us would not be the same.

Figure 1: Intermolecular and Intramolecular Forces



Chemical Thinking Questions

- What property influences the experimental results shown in Figure 1?
A. Intermolecular forces
B. Intramolecular forces
- On what factor does the molecule B, where H_2O -like intermolecular forces exist, Figure 1, mainly depend to increase this force?
A. Volume and shape
B. Intensity of bonds
- Based on the intermolecular forces and molecule B, does water molecules (H_2O) have intermolecular forces similar to molecule B and C?
A. Yes
B. No
- On what factor another molecule, where intermolecular forces tend to cause molecules the tendency to associate with each other? Explain. The hydroxyl group.
A. Intensity of bonds
B. Volume and shape
- When intermolecular forces exist in Figure 1, where are they generated and how do the molecules — with the intermolecular forces between the molecules? Explain fully. (10 points)
A. Between atoms
B. Between molecules
- Does the difference between intermolecular and intramolecular forces become evident when water is the molecule itself?
A. Between atoms
B. Between molecules
C. Between molecules and atoms
D. Between atoms and molecules

Pogil Intermolecular Forces

H. Margenau,N. R. Kestner

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Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog,2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes

Making Chemistry Relevant Sharmistha

Basu-Dutt,2010-02-19 Unique new approaches for making chemistry accessible to diverse students Students interest and achievement in academics improve dramatically when they make connections between what they are learning and the potential uses of that knowledge i n the workplace and or in the world at large Making Chemistry Relevant presents a unique collection of strategies that have been used successfully in chemistry classrooms to create a learner sensitive environment that enhances academic achievement and social competence of students Rejecting rote memorization the book proposes a cognitive constructivist philosophy that casts the teacher as a facilitator helping students to construct solutions to problems Written by chemistry professors and research groups from a wide variety of colleges and universities the book offers a number of creative ways to make chemistry relevant to the student including Teaching science in the context of major life issues and STEM professions Relating chemistry to current events such as global warming pollution and terrorism

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Analytical Chemistry Juliette Lantz,Renée Cole,The POGIL Project,2014-12-31 An essential guide to inquiry approach instrumental analysis

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Chemists' Guide to

Effective Teaching Norbert J. Pienta,Melanie M. Cooper,Thomas J. Greenbowe,2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry

Intermolecular Forces Geoffrey C. Maitland,1981

Theory of Intermolecular Forces H. Margenau,N. R. Kestner,2013-10-22 Theory of Intermolecular Forces deals with the exposition of the principles and techniques of the theory of intermolecular forces The text focuses on the basic theory and

surveys other aspects with particular attention to relevant experiments The initial chapters introduce the reader to the history of intermolecular forces Succeeding chapters present topics on short intermediate and long range atomic interactions properties of Coulomb interactions shape dependent forces between molecules and physical adsorption The book will be of good use to experts and students of quantum mechanics and advanced physical chemistry

Intermolecular Forces Pierre

L. Huyskens,Werner A.P. Luck,Therese Zeegers-Huyskens,2012-12-06 The study of intermolecular forces began over one hundred years ago in 1873 with the famous thesis of van der Waals In recent decades knowledge of this field has expanded due to intensive research into both its theoretical and the experimental aspects This is particularly true for the type of very strong cohesive force stressed in 1920 by Latimer and Rodebush the hydrogen bond a phenomenon already outlined in 1912 by Moore and Winemill Hydrogen bonds exert a profound influence on most of the physical and chemical properties of the materials in which they are formed Not only do they govern viscosity and electrical conductivity they also intervene in the chemical reaction path which determines the kinetics of chemical processes The properties of chemical substances depend to a large extent on intermolecular forces In spite of this fundamental fact too little attention is given to these properties both in research and in university teaching For instance in the field of pharmaceutical research about 13000 compounds need to be studied in order to find a single new product that can be successfully marketed The recognition of the need to optimize industrial research efficiency has led to a growing interest in promoting the study of inter molecular forces Rising salary costs in industry have encouraged an interest in theoretical ideas which will lead to tailor made materials

Theory of Intermolecular Forces Henry Margenau,N. R. Kestner,1969

Intermolecular and Surface Forces Jacob N.

Israelachvili,2010-12-02 This reference describes the role of various intermolecular and interparticle forces in determining the properties of simple systems such as gases liquids and solids with a special focus on more complex colloidal polymeric and biological systems The book provides a thorough foundation in theories and concepts of intermolecular forces allowing researchers and students to recognize which forces are important in any particular system as well as how to control these forces This third edition is expanded into three sections and contains five new chapters over the previous edition Starts from the basics and builds up to more complex systems Covers all aspects of intermolecular and interparticle forces both at the fundamental and applied levels Multidisciplinary approach bringing together and unifying phenomena from different fields This new edition has an expanded Part III and new chapters on non equilibrium dynamic interactions and tribology friction forces

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