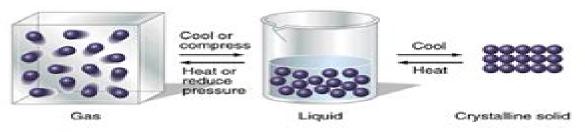
## POGIL: Kinetic Molecular Theory

#### Learning Objectives

- Identify the basic differences between particle behavior in the solid, liquid, and gaseous phases.
- Develop an understanding of the postulates of the kinetic molecular theory.
- Identify how temperature affects molecular motion.
- Apply the kinetic molecular theory to predict the outcome of everyday situations.

#### Model 1: Representation of Atoms in Different Phases

The following figures show the representative particles of a given substance in various states of matter.



Source: http://itl.chem.ufl.edu/2045\_s00/lectures/lec\_f.html

Note: Give complete answers to the key questions, exercises, and problems. Provide explanations and show work, where necessary.

#### Key Questions

 What are the key characteristics of atoms and molecules in gases, liquids, and solids? In Table I below, describe the characteristics of particles for each phase of matter based on Model 1. Be specific with regard to spacing, the <u>potential</u> of particles movement, and whether or not the particles will fill the container.

Table 1: Characteristics of the Phases of Matter

	SOLID	LIQUID	GAS
SPACING AND RELATIVE DENSITY			
MOVEMENT & RELATIVE KINETIC ENERGY			
FILLING A CONTAINER (VOLUME)			
"SHAPE" AND MASS AND MISC			

# **Pogil Kinetic Molecular Activity Answers**

Charles Proteus Steinmetz, American Electrochemical Society. Meeting

# **Pogil Kinetic Molecular Activity Answers:**

Kinetic Molecular Theory Charles Proteus Steinmetz, American Electrochemical Society. Meeting, 1908\*

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