

## Model 2 – Ternary Acids (Oxyacids)

Acid	Name of Acid in Aqueous Solution	Cation (+)	Polyatomic Anion (-)	Polyatomic Anion Name
$\text{HClO}_3$	Chloric acid	$\text{H}_3\text{O}^+$	$\text{ClO}_3^{-1}$	Chlorate
$\text{H}_2\text{SO}_3$	Sulfurous acid	$2\text{H}_3\text{O}^+$	$\text{SO}_3^{-2}$	Sulfite
$\text{H}_2\text{SO}_4$	Sulfuric acid	$2\text{H}_3\text{O}^+$	$\text{SO}_4^{-2}$	Sulfate
$\text{H}_3\text{PO}_3$	Phosphorous acid	$3\text{H}_3\text{O}^+$	$\text{PO}_3^{-3}$	Phosphite
$\text{H}_3\text{PO}_4$	Phosphoric acid	$3\text{H}_3\text{O}^+$	$\text{PO}_4^{-3}$	Phosphate
$\text{HNO}_3$	Nitric acid	$\text{H}_3\text{O}^+$	$\text{NO}_3^{-1}$	Nitrate
$\text{HNO}_2$	Nitrous acid	$\text{H}_3\text{O}^+$	$\text{NO}_2^{-1}$	Nitrite
$\text{H}_2\text{CO}_3$	Carbonic acid	$2\text{H}_3\text{O}^+$	$\text{CO}_3^{-2}$	Carbonate

6. Look at the formulas of the ternary acids in Model 2.

a. How are ternary acids different from binary acids in their structure?

*Ternary acids contain three elements and they also contain the element oxygen.*

b. What number do you think the prefix "ter-" refers to?

*"ter" refers to three*

7. When ternary acids are mixed with water, ions will form. Fill in the table above with the formulas and names of the anions.

8. Examine the pairs of ternary acids in Model 2 that contain sulfur, phosphorus, and nitrogen. Each pair has one acid that ends in "-ic" and another that ends in "-ous." These endings are related to the name of the polyatomic anion found in the acid ("-ate" or "-ite"). Complete the statements below with the correct acid name ending.

Polyatomic anion ending is "-ate" → acid name ending is "ic"

Polyatomic anion ending is "-ite" → acid name ending is "ous"

9. If the prefix "hydro-" were used to name a ternary acid, what problem would this create when naming  $\text{HClO}_3$ ?

*$\text{ClO}_3^{-1}$  has the name of Chlorate. "ate" must be replaced with "ic". If "hydro" is used as a prefix, it would be hydrochloric acid. This has the formula  $\text{HCl}$ , not  $\text{HClO}_3$ .*

10. Write a rule for naming ternary acids.

*- The prefix "hydro" is not used in the name when an acid contains more than two elements and one of the elements is oxygen.*

*- "ate" is replaced with "ic" and "ite" is replaced with "ous".*

11. Predict the formula for chlorous acid.



12. Circle the acid(s) below that would be named beginning with the prefix "hydro-."



# Pogil Solubility Answer Sheet

**Jin-Ying Zhang**

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## **Table of Contents Pogil Solubility Answer Sheet**

1. Understanding the eBook Pogil Solubility Answer Sheet
  - The Rise of Digital Reading Pogil Solubility Answer Sheet
  - Advantages of eBooks Over Traditional Books
2. Identifying Pogil Solubility Answer Sheet
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Pogil Solubility Answer Sheet
  - User-Friendly Interface
4. Exploring eBook Recommendations from Pogil Solubility Answer Sheet
  - Personalized Recommendations
  - Pogil Solubility Answer Sheet User Reviews and Ratings
  - Pogil Solubility Answer Sheet and Bestseller Lists
5. Accessing Pogil Solubility Answer Sheet Free and Paid eBooks

- Pogil Solubility Answer Sheet Public Domain eBooks
- Pogil Solubility Answer Sheet eBook Subscription Services
- Pogil Solubility Answer Sheet Budget-Friendly Options
- 6. Navigating Pogil Solubility Answer Sheet eBook Formats
  - ePub, PDF, MOBI, and More
  - Pogil Solubility Answer Sheet Compatibility with Devices
  - Pogil Solubility Answer Sheet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Pogil Solubility Answer Sheet
  - Highlighting and Note-Taking Pogil Solubility Answer Sheet
  - Interactive Elements Pogil Solubility Answer Sheet
- 8. Staying Engaged with Pogil Solubility Answer Sheet
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Pogil Solubility Answer Sheet
- 9. Balancing eBooks and Physical Books Pogil Solubility Answer Sheet
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Pogil Solubility Answer Sheet
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Pogil Solubility Answer Sheet
  - Setting Reading Goals Pogil Solubility Answer Sheet
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pogil Solubility Answer Sheet
  - Fact-Checking eBook Content of Pogil Solubility Answer Sheet
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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