

Acids and Bases

How do acids and bases behave in water?

Why?

Acids and bases play an important role in our lives. Numerous biological processes, industrial applications, and even environmental problems are a function of the acidity or basicity (alkalinity) of aqueous solutions. It is therefore important to understand what makes a substance behave as an acid or a base when dissolved in water. In this activity, we will explore the physical and chemical properties of acids and bases.

Model 1 – Arrhenius Acids and Bases

| Common Name for Aqueous Solution | Chemical Formula | Found in... | Tastes... | Turns Litmus Paper... | Conducts Electricity? | Acid or Base |
|----------------------------------|-------------------------------------------------------|-------------------|-----------|-----------------------|-----------------------|--------------|
| Acetic acid | $\text{HC}_2\text{H}_3\text{O}_2(\text{aq})$ | Vinegar | Sour | Red | Yes | Acid |
| Benzoic acid | $\text{HC}_7\text{H}_5\text{O}_2(\text{aq})$ | Food preservative | Sour | Red | Yes | Acid |
| Phosphoric acid | $\text{H}_3\text{PO}_4(\text{aq})$ | Soda pop | Sour | Red | Yes | Acid |
| Hydrochloric acid | $\text{HCl}(\text{aq})$ | Stomach acid | Sour | Red | Yes | Acid |
| Citric acid | $\text{H}_3\text{C}_6\text{H}_5\text{O}_7(\text{aq})$ | Citrus fruits | Sour | Red | Yes | Acid |
| Ascorbic acid | $\text{H}_2\text{C}_6\text{H}_8\text{O}_6(\text{aq})$ | Vitamin C | Sour | Red | Yes | Acid |
| Magnesium hydroxide | $\text{Mg}(\text{OH})_2(\text{aq})$ | Milk of magnesia | Bitter | Blue | Yes | Base |
| Aluminum hydroxide | $\text{Al}(\text{OH})_3(\text{aq})$ | Antacids | Bitter | Blue | Yes | Base |
| Barium hydroxide | $\text{Ba}(\text{OH})_2(\text{aq})$ | Lubricants | POISON | Blue | Yes | Base |
| Sodium hydroxide | $\text{NaOH}(\text{aq})$ | Drain cleaner | POISON | Blue | Yes | Base |

1. Refer to Model 1.
 - a. What is the common chemical name for Vitamin C?
 - b. Is Vitamin C classified as an acid or a base?
2. Examine the properties of the Arrhenius acids in Model 1. List three properties that all Arrhenius acids have in common.
3. Examine the chemical formulas for the Arrhenius acids in Model 1. What feature do all the Arrhenius acid chemical formulas have in common?
4. Examine the properties of the Arrhenius bases in Model 1. List two properties that all Arrhenius bases have in common.

Pogil Unit 3acids And Bases Answer Key

RJ Shavelson



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