

Polar Bears and Penguins



Directions: Read the picture above to answer the following questions.

1. Check the Lewis Dot Structure for HCl. Place drawing below.



2. If that penguins represents the hydrogen atom and the polar bear represents the chlorine atom, what does the two dots represent in that drawing?

a. It is the bonded pair of electrons (chlorine, hydrogen)

3. What do the polar bears represent in trying to illustrate?

a. Electronegativity between atoms

Directions: Read the [Polar Bear and Penguin Analogy](#), answer the following questions as you read.

1. In polar bear analogy, what electronegativity?

a. Ability of an atom to attract bonding electrons towards itself

2. Polar bears like to eat. What do they eat polar molecules?

a. A molecule with a difference in electrical charge

3. Fill out the table below to answer the following questions, what does hydrogen and chlorine eat represented in polar bear analogy? What happens in that bonding, electrons are more likely to be with?

Type of Bond	What happens to the bonding electrons in each type of bond?
Nonpolar covalent	The electrons do not spend more time around either of the atoms.

Polar Bears And Penguins Chemistry Answers

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