Exercise 1: Qualitative Anion Tests

Data Table 1. Anion Confirmation Tests.

Chemical	Anion	Addition of AgNO ₃	Addition of HCI
NaBr	Bromide	Milky like precipitate formed	No initial changes, solids began to fall out after an hour
Na ₂ CO ₃	Carbonate	Precipitate formed, white in color, and began to separate	Mixture cleared, precipitate clumped together, small bubble could be seen on the bottom
NaCl	Chloride	Mixture turned completely white, small amount of clumping from precipitate being formed	Solids pulled out of the mixture; no other reaction occurred
Nal	lodide	Mixture turned opalescent; precipitate formed	Colors in mixture faded slightly, no other reaction
Na ₂ SO ₄	Sulfate	No observable reaction	Mixture turned white
Na₂S	Sulfide	Mixture turned brown black in color, solids forms and sank to bottom.	No noticeable changes, did notice smell of rotten eggs,
Unknown #103	Carbonate	Solids pulled out of mixture	Bubble formed
Unknown #104	Sulfide	Turned brown black, solids separated and sank	Rotten eggs smell

Qualitative Anion Tests Lab Answers

Michael R. Abraham, Michael J. Pavelich

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<u>CliffsNotes AP Chemistry</u> Bobrow Test Preparation Services,2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter s end Two full length practice tests with detailed answer explanations are included in the book

Laboratory Manual for Principles of General Chemistry Jo Allan Beran, 2010-11-01 This new edition of the Beran lab manual emphasizes chemical principles as well as techniques. The manual helps students understand the timing and situations for the various techniques The Beran lab manual has long been a market leading lab manual for general chemistry Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by Laboratory Manual for Principles of General Chemistry J. A. Beran, Mark Lassiter, 2022-08-16 Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure selecting the proper apparatus employing the proper techniques systematically analyzing and interpreting the data and minimizing inherent variables As a result of good data a scientific and analytical conclusion is made which may or may not be right but is certainly consistent with the data Experiments write textbooks textbooks don t write experiments A student's scientific literacy grows when experiences and observations associated with the scientific method are encountered Further experimentation provides additional cause effect observations leading to an even better understanding of the experiment The 11th edition s experiments are informative and challenging while offering a solid foundation for technique safety and experimental procedure The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit An Additional Notes column is included in each experiment's Report Sheet to provide a space for recording observations and data during the experiment Continued emphasis on handling data is supported by the Data Analysis section Laboratory Course Chemistry Mr. Rohit Manglik, 2024-03-15 EduGorilla Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels

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