

## 36 – Quantum Mechanics – The Photoelectric Effect

**Purpose:** Investigate the interaction of photons with metals and the resulting ejection of high-energy electrons.

**Apparatus:** PhET Simulation-Photoelectric Effect

PhET Simulations → Play With Sims → Quantum Phenomena → Photoelectric Effect

Rank 10/10

**Discussion:** That light can diffract and refract tells us light behaves like a wave. How do we know that light behaves like a particle? The photoelectric effect leads us to this. How does your solar-powered calculator work? Solar PV cells (photovoltaic) convert the energy in light to electric energy (DC current). The same principles that allow a solar-powered calculator to work also act in your cell phone's digital camera. For those materials that we classify as photosensitive (most metals, metalloids), light comes in, electrons come out. If we reverse the process...Lasers...



Photoelectric Effect

**Important Formulas and Constants:**  $E = hf$   $E = pc$   $c = f\lambda$

$$KE_{\text{max}} = hf - \phi \quad \phi = hf_0 \quad 1.00\text{eV} = 1.60 \times 10^{-19}\text{C}V = 1.60 \times 10^{-19}\text{J} \quad h = 6.63 \times 10^{-34}\text{Js} \quad 1V = 1J/1C$$

**Procedure, Part I: Determining the Energy threshold electrons.**

1. Take some time and familiarize yourself with the excellent simulation. You can change the photosensitive metal observed, the light intensity (brightness), voltage added, and the very important EM wavelength.
2. Remember...wavelength is used in the simulation, and frequency is used in the formulas. The product of the two is the speed of light,  $3.00 \times 10^8\text{ m/s}$ .  $c = f\lambda$
3. Click on the boxes to show the graphs of  $I$  vs  $V$ ,  $I$  vs  $A$ , and  $E$  vs  $f$
4. Using **sodium, Na** as our metal, change the wavelength to a value that just begins to cause current to flow.



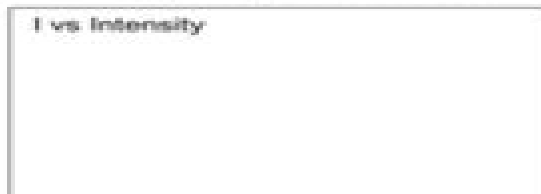
**Threshold wavelength** required eject electrons from sodium: \_\_\_\_\_ nm

**Threshold frequency** of this wavelength: \_\_\_\_\_ Hz

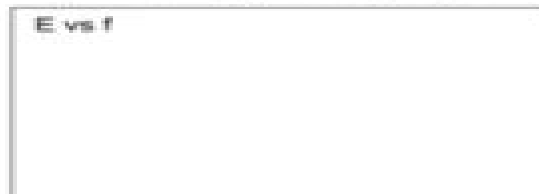
**Minimum energy** each photon must contain: \_\_\_\_\_ J and \_\_\_\_\_ eV

5. Next, set the wavelength to 400 nm and intensity to 50% record the current flow here: \_\_\_\_\_ A
6. What effect does increasing the intensity (at this wavelength) have on the current flow?  
\_\_\_\_\_
7. Set the wavelength to 700nm. What effect does increasing the intensity (at this wavelength) have on the current flow? \_\_\_\_\_
8. Why is this? \_\_\_\_\_
9. Sketch sample graphs of current as intensity changes and energy and frequency changes:

I vs Intensity



E vs f



# Quantum Mechanics The Photoelectric Effect Phet Answers

**Puja Mehta**



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