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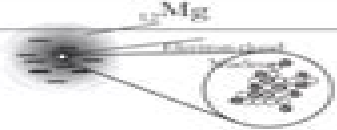
Isotopes

Are all atoms of an element alike?

Why?

The following activity will help you learn the important structural characteristics of an atom. How do we classify atoms? How does the combination of subatomic particles affect the mass and charge of an atom? What are isotopes? This is just a sampling of what we will address. Throughout this activity you will want to keep both Model 1 and a periodic table handy.

Model 1

Isotopes of Hydrogen			
Isotope Symbol	${}^1_1\text{H}$	${}^2_1\text{H}$	${}^3_1\text{H}$
Atomic Diagram with Name			
Number of Protons	1		
Number of Neutrons	0		
Isotopes of Carbon			
Isotope Symbol	${}^{12}_6\text{C}$	${}^{13}_6\text{C}$	${}^{14}_6\text{C}$
Atomic Diagram with Name			
Number of Protons	6		
Number of Neutrons	6		
Isotopes of Magnesium			
Isotope Symbol	${}^{24}_{12}\text{Mg}$	${}^{25}_{12}\text{Mg}$	${}^{26}_{12}\text{Mg}$
Atomic Diagram with Name			
Number of Protons	12		
Number of Neutrons	12		

Pogil Chapter 11 Key

Christian Drosten

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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