



## Extension Questions

### Model 2 – Combustion Reactions



( $\text{C}_8\text{H}_{18}$  = octane—gasoline component)



( $\text{C}_4\text{H}_{10}$  = butane—lighter fuel)



( $\text{CH}_4$  = methane—natural gas)

19. What are the products in all of the combustion reactions in Model 2?

$\text{CO}_2$  and  $\text{H}_2\text{O}$

20. What reactant is common in all of the combustion reactions in Model 2?

$\text{O}_2$

21. The "fuel" in most combustion reactions is a **hydrocarbon**. Using the examples in Model 2, write a description for the classification of substances known as hydrocarbons.

compound containing H + C

22. Predict the products of the following combustion reactions, and balance each reaction equation.



23. Compounds such as isopropyl alcohol ( $\text{C}_3\text{H}_7\text{OH}$ ) and glucose ( $\text{C}_6\text{H}_{12}\text{O}_6$ ) can also undergo combustion in a similar way. Predict the products of the following combustion reactions, and balance each reaction equation.



24. For each of the reaction types in Model 1, explain why a combustion reaction CANNOT be classified as that type. Use a complete sentence in each answer.

S → combustion reactions don't just have 1 product

D → "

" have more than 1 reactant

SR → "

" need  $\text{O}_2$  in both products not just 1

DR → "

" start w/ a compound and an element  
not 2 compounds

# Pogil Activities For High School Chemistry Answers Extension Questions

**M Walker**



## **Pogil Activities For High School Chemistry Answers Extension Questions:**

## **Pogil Activities For High School Chemistry Answers Extension Questions** Book Review: Unveiling the Power of Words

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