

- Which illustration below represents
 - solute particles in a solid state in water?
 - solute particles in an aqueous state?



- What variables are controlled in all five beakers of Model 1?

Same amount of solvent(water), same temperature, same solute, stirred for 2 hours

- Count the particles present in each beaker of Model 1. Fill in the table to show the number of dissolved solute particles and the number of solid solute particles.

Refer to table.

- Consider the beakers in Model 1.
 - Which beakers represent **unsaturated** solutions?
A & B
 - Which beakers represent **saturated** solutions?
C, D & E
- Beakers A–E in Model 1 are depicted as representing five different or separate solutions. They could also be considered as five “snapshots” of the same beaker over time. In other words, if additional measured quantities of solute were stirred into beaker A in small increments over time, then beakers B–E would result.
 - When a small amount of additional solute is added to an unsaturated solution, what happens to the number of dissolved particles? Provide specific evidence from Model 1 to support your answer.

The particles dissolve. If we look at beaker A and we add more solute it becomes like beaker B which has more particles dissolved.
 - When a small amount of additional solute is added to a saturated solution, what happens to the number of dissolved particles? Provide specific evidence from Model 1 to support your answer.

The particles settle out to the bottom, this is like going from beaker C to D.
 - Predict what would happen if a small amount of additional solute were stirred into beaker E in Model 1.

Nothing would happen, more solute would settle to the bottom.
- Have each person in your group provide an example of the word “saturated” as it is used in an everyday context. Summarize the meaning of the word in the space below.

Some examples: a soaked sponge, soil after a heavy rain, adding sugar to iced tea
Definition: something that has the maximum amount of a substance in it.

Pogil Aqueous Solutions Answers

Robert Griffin Smith



Pogil Aqueous Solutions Answers:

Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes

Analytical Chemistry Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the BerriLambert atomic and molecular absorption processes vibrational modes mass spectra interpretation and much more

Aqueous Solutions Rolf Kurt Freier, 1976

Water And Aqueous Solutions Neilson G., Enderby J E., 1976

Elementary Study of Aqueous Solutions Harold Wilberforce Close, 1941

Introductory Chemistry Michael P. Garoutte, Ashley B. Mahoney, 2015-08-10 The ChemActivities found in Introductory Chemistry A Guided Inquiry use the classroom guided inquiry approach and provide an excellent accompaniment to any one semester Introductory text Designed to support Process Oriented Guided Inquiry Learning POGIL these materials provide a variety of ways to promote a student focused active classroom that range from cooperative learning to active student participation in a more traditional setting

Supplements / Ergänzungen Rolf K. Freier, 2011-06-01 No detailed description available for Supplements Erg n zungen

The Osmotic Pressure of Aqueous Solutions, 1914

Aqueous Solutions Lifelique, 2019 This lesson plan covers the definition of a solution and identifies its parts how aqueous solutions are created from both ionic and molecular compounds explains insolubility and distinguishes between electrolytes and non electrolytes

The Structure of Aqueous Solutions Alan K. Soper, 1977

Hydrates in Aqueous Solution Frederick Hutton Getman, Harry Clary Jones, Harry Preston Bassett, 2016-05-22 This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it This work was reproduced from the original artifact and remains as true to the original work as possible Therefore you will see the original copyright references library stamps as most of these works have been housed in our most important libraries around the world and other notations in the work This work is in the public domain in the United States of America and possibly other nations Within the United States you may freely copy and distribute this work as no entity individual or corporate has a copyright on the body of the work As a reproduction of a historical artifact this work may contain missing or blurred pages poor pictures errant marks etc Scholars believe and we concur that this work is important enough to be preserved reproduced and made generally available to the public We appreciate your support of the preservation process and thank you for being an important part of keeping this knowledge alive and relevant

Liquid Water and Aqueous Solutions Alan K. Soper, 2000

HYDRATES IN AQUEOUS SOLUTION E Harry Clary 1865-1916 Jones, 2016-08-26 This work has been

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A Study of the Solubility of Certain Organic Substances in Aqueous Solutions John Henry Saylor, 1930 A Study of the Free Discharge of Non-aqueous Solutions Walter Carrington Fitzpatrick, 1938

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