

# Empirical and Molecular Formulas Practice

Read the question carefully for the data provided. Answers must be shown for credit.

Find empirical formula:

1. A compound is found to contain 40.94 g of iron and 35.93 g of sulfur. Find the empirical formula.

$$\text{40.94 g Fe} \times \frac{1 \text{ mol Fe}}{55.85 \text{ g Fe}} = 0.7332 \text{ mol Fe}$$

$$\text{35.93 g S} \times \frac{1 \text{ mol S}}{32.06 \text{ g S}} = 1.121 \text{ mol S}$$

FeS

2. A compound is found to contain 26.88 g of potassium, 35.98 g of chlorine, and 37.14 g of oxygen. Find the empirical formula.

$$\text{26.88 g K} \times \frac{1 \text{ mol K}}{39.10 \text{ g K}} = 0.6875 \text{ mol K}$$

$$\text{35.98 g Cl} \times \frac{1 \text{ mol Cl}}{35.45 \text{ g Cl}} = 1.015 \text{ mol Cl}$$



$$\text{37.14 g O} \times \frac{1 \text{ mol O}}{16.00 \text{ g O}} = 2.321 \text{ mol O} \left( \frac{2.321 \text{ mol O}}{0.6875} = 3.376 \text{ O} \approx 3 \right)$$

3. A 10.00 g sample of a compound, a gas, contains 2.74 g of carbon, 4.34 g of hydrogen, and 2.92 g of oxygen. Find the empirical formula.

$$\text{2.74 g C} \times \frac{1 \text{ mol C}}{12.01 \text{ g C}} = 0.2281 \text{ mol C}$$



$$\text{4.34 g H} \times \frac{1 \text{ mol H}}{1.008 \text{ g H}} = 4.305 \text{ mol H} \left( \frac{4.305 \text{ mol H}}{0.2281} = 18.87 \text{ H} \approx 19 \right)$$

$$\text{2.92 g O} \times \frac{1 \text{ mol O}}{16.00 \text{ g O}} = 0.1825 \text{ mol O} \left( \frac{0.1825 \text{ mol O}}{0.2281} = 0.800 \text{ O} \approx 1 \right)$$

4. A 25.00 g sample of an ionic compound contains 29.88 g sodium, 43.48 g chlorine, and 26.64 g oxygen. Find the empirical formula.

$$\text{29.88 g Na} \times \frac{1 \text{ mol Na}}{22.99 \text{ g Na}} = 1.299 \text{ mol Na}$$



$$\text{43.48 g Cl} \times \frac{1 \text{ mol Cl}}{35.45 \text{ g Cl}} = 1.226 \text{ mol Cl}$$

$$\text{26.64 g O} \times \frac{1 \text{ mol O}}{16.00 \text{ g O}} = 1.665 \text{ mol O} \left( \frac{1.665 \text{ mol O}}{1.299} = 1.282 \text{ O} \approx 1.3 \text{ O} \approx \frac{13}{10} \right)$$

# Pogil Chemistry Answer Key Empirical

**R Barnett**



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