

Bond Energy

What makes a reaction endothermic or exothermic?

Why?

Chemical reactions can be a lot like playing with Legos®—you must take apart part of your last creation before you can replace it with something new. For many chemical reactions, we have to first break bonds in the reactants before we can put the atoms back into a new arrangement to form the products. Both of these processes involve changes in energy. The net energy change for a reaction is called the **heat of reaction** or the **change in enthalpy** (ΔH). In this activity we will look at one way energy changes can be approximated for chemical reactions.

Model 1 – Breaking and Forming Bonds

	Reaction	Change in Energy (kJ/mol _{rxn})		Reaction	Change in Energy (kJ/mol _{rxn})
A)	$\text{PCl}_3(\text{g}) \rightarrow \text{P}(\text{g}) + 3\text{Cl}(\text{g})$	+ 966.7 <i>broken</i>	E)	$\text{P}(\text{g}) + 3\text{Cl}(\text{g}) \rightarrow \text{PCl}_3(\text{g})$	- 966.7 <i>formed</i>
B)	$\text{PCl}_5(\text{g}) \rightarrow \text{P}(\text{g}) + 5\text{Cl}(\text{g})$	+ 1,297.9 <i>broken</i>	F)	$\text{P}(\text{g}) + 5\text{Cl}(\text{g}) \rightarrow \text{PCl}_5(\text{g})$	- 1,297.9 <i>formed</i>
C)	$\text{PF}_3(\text{g}) \rightarrow \text{P}(\text{g}) + 3\text{F}(\text{g})$	+ 1,470.4 <i>broken</i>	G)	$\text{P}(\text{g}) + 3\text{F}(\text{g}) \rightarrow \text{PF}_3(\text{g})$	- 1,470.4 <i>formed</i>
D)	$\text{PF}_5(\text{g}) \rightarrow \text{P}(\text{g}) + 5\text{F}(\text{g})$	+ 2,305.4 <i>broken</i>	H)	$\text{P}(\text{g}) + 5\text{F}(\text{g}) \rightarrow \text{PF}_5(\text{g})$	- 2,305.4 <i>formed</i>

- Identify four reactions in Model 1 where bonds are being broken. Write "broken" in the Change in Energy box in the table for each reaction you identify.
- Identify four reactions in Model 1 where bonds are being formed. Write "formed" in the Change in Energy box in the table for each reaction you identify.
- Circle the correct word to complete each sentence below.
 - When bonds are (broken/formed) there is a positive energy change.
 - Breaking bonds is (endothermic/exothermic).
 - When bonds are (broken/formed) there is a negative energy change.
 - Forming bonds is (endothermic/exothermic).
- Find two reactions in Model 1 that are exact opposites of each other, that is, one reaction is the reverse of the other reaction.

- How do the changes in energy for the reverse reactions compare?

The changes in energy for these reactions are the same, but opposite in sign.

- Explain your answer to part a considering what you learned from Questions 1–3 about bond breaking and bond formation.

If a given amount of energy is required to break a bond, then the same amount of energy will be released when the bond is formed.

Pogil Energy Answers

Marco Cascella

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

Pogil Energy Answers:

Organic Chemistry Suzanne M. Ruder, The POGIL Project, 2015-12-29 ORGANIC CHEMISTRY **POGIL** Shawn R. Simonson, 2023-07-03 Process Oriented Guided Inquiry Learning POGIL is a pedagogy that is based on research on how people learn and has been shown to lead to better student outcomes in many contexts and in a variety of academic disciplines Beyond facilitating students mastery of a discipline it promotes vital educational outcomes such as communication skills and critical thinking Its active international community of practitioners provides accessible educational development and support for anyone developing related courses Having started as a process developed by a group of chemistry professors focused on helping their students better grasp the concepts of general chemistry The POGIL Project has grown into a dynamic organization of committed instructors who help each other transform classrooms and improve student success develop curricular materials to assist this process conduct research expanding what is known about learning and teaching and provide professional development and collegiality from elementary teachers to college professors As a pedagogy it has been shown to be effective in a variety of content areas and at different educational levels This is an introduction to the process and the community Every POGIL classroom is different and is a reflection of the uniqueness of the particular context the institution department physical space student body and instructor but follows a common structure in which students work cooperatively in self managed small groups of three or four The group work is focused on activities that are carefully designed and scaffolded to enable students to develop important concepts or to deepen and refine their understanding of those ideas or concepts for themselves based entirely on data provided in class not on prior reading of the textbook or other introduction to the topic The learning environment is structured to support the development of process skills such as teamwork effective communication information processing problem solving and critical thinking The instructor s role is to facilitate the development of student concepts and process skills not to simply deliver content to the students The first part of this book introduces the theoretical and philosophical foundations of POGIL pedagogy and summarizes the literature demonstrating its efficacy The second part of the book focusses on implementing POGIL covering the formation and effective management of student teams offering guidance on the selection and writing of POGIL activities as well as on facilitation teaching large classes and assessment The book concludes with examples of implementation in STEM and non STEM disciplines as well as guidance on how to get started Appendices provide additional resources and information about The POGIL Project **Analytical Chemistry** Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the BerriLambert atomic and molecular absorption processes vibrational modes

mass spectra interpretation and much more *Argumentation in Chemistry Education* Sibel Erduran, 2022-06-29 Scientists use arguments to relate the evidence that they select from their investigations and to justify the claims that they make about their observations This book brings together leading researchers to draw attention to research policy and practice around the inclusion of argumentation in chemistry education **Advances in Teaching Physical Chemistry** Mark David

Ellison, 2008 This book brings together the latest perspectives and ideas on teaching modern physical chemistry It includes perspectives from experienced and well known physical chemists a thorough review of the education literature pertaining to physical chemistry a thorough review of advances in undergraduate laboratory experiments from the past decade in depth descriptions of using computers to aid student learning and innovative ideas for teaching the fundamentals of physical chemistry This book will provide valuable insight and information to all teachers of physical chemistry **Answers to Your**

Energy Questions Canada. Department of Energy, Mines and Resources, 1978 *Answers to Your Energy Question*

Canada. Office of Energy Conservation, 1977 **Energy - the Task Ahead - Policy Questions and Answers**, 1977

ANSWERS TO YOUR ENERGY QUESTIONS. Canada. Energy Policy Sector, 1977 *Chemical Energy*, 1991

We've Got All the Answers to Your Home Energy Questions Under One Roof Alberta. Energy Conservation Branch, Canada. Energy, Mines and Resources Canada, 1986 **The Rough Guide to the Energy Crisis** David Buchan, 2012-06-21 Climate Change is forcing us to re think our use of fossil fuels the oil coal and gas that we have depended upon for generations A rising global population means that there is an unprecedented level of demand on the world's energy resources And there are still desperately poor areas of the world that remain unconnected to a national grid The Rough Guide to the Energy Crisis examines these important issues and explains the many challenges facing energy today and explores possible solutions KEY TOPICS INCLUDE Energy today We take our energy for granted But what do we use and where does it come from Peak Oil How long do we have before oil production peaks Has it peaked already Renewables Can wind turbines solar panels and wave and tidal power really keep the lights on Nuclear How safe are the power plants And can you be a pro nuclear green Climate Change Our energy use has environmental consequences Are their technological answers And can our lifestyles adapt Keeping the planet safe while keeping the world moving is perhaps the most important challenge we currently face The Rough Guide to the Energy Crisis examines the many sides of this problem and how it could be approached with erudition and accessibility **370 Computerized Questions & Answers about Energy** Hassan

Rafi-Zadeh, 1974 **Energy Research** Edison Electric Institute, 1978 Energy Joseph Priest, 1991 The fourth edition of Joseph Priest's *Energy* draws heavily on the experience of instructors who used earlier editions It is designed for environmental science courses for non science majors and is suitable for fulfilling their science requirement The amount of quantitative material exposes students to the importance of quantitative reasoning without causing math anxiety Priest has refined the physics and integrated it with energy issues added new topics updated the energy data and included new

exercises and problems The fourth edition includes coverage of acid precipitation the greenhouse effect depletion of the ozone layer and radon radioactivity in houses The text furthers students understanding with references that provide up to date sources of information and a guide to the types of publications and books students should examine review sections with questions highlighting the main topics and ideas in the chapters as well as 25 multiple choice questions and answers and questions and problems sections that emphasize written answers and practical numerical calculations

Energy and Us Glenn A. Gibson, 2011-04-01 Energy and Us opens with an overview of energy and is followed with a detailed analysis of the production consumption uses reserves costs advantages and disadvantages of petroleum natural gas coal and uranium which are nonrenewable fuels A similar evaluation covers biomass geothermal tidal solar water and wind which are renewable sources of energy The readers then learn about electricity its uses and how it is generated The Energy Information Administration in the Department of Energy divides our energy usage into residential commercial industrial and transportation sectors and Glenn Gibson examines these before discussing the pollution associated with energy global warming and the environment in general Finally he presents an historic review of energy policy While not overly technical this book does require some knowledge of elementary chemistry to fully appreciate the data and so the history and science of energy is provided in an appendix for those who would like a refresher course A summary of the units of measure and the conversion between them is also included along with a list of acronyms and their meanings Although the author offers possible solutions to problems these solutions are primarily based on data obtained from the Energy Information Administration and Bureau of Transportation Statistics in the Department of Transportation It is hoped that readers will assimilate the information says Gibson and decide for themselves what we should do He believes it is predominantly our resistance to change not conservation or the lack of technology that will determine our energy future and our consequent standard of living It is unquestionable that many of our existing energy sources are limited and this illuminating and fact filled book is sure to increase awareness of how we use energy to maintain our current living standards and why we need to change the way we actually acquire that energy as soon as possible

Questions and Answers on the Energy Crisis Institute of Human Relations (American Jewish Committee), 1973 Issues & Answers, 1981 **Energy** Glen Phelan, 2008

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