

Phases and Intermolecular Forces

Each of the following sets of molecules are similar in either Electron Pair Geometry or Molecular Geometry. Even when the shapes are the same the polarity may vary. Identify shape and polarity for each structure, and then consider the questions posed.

1. There are several ways to be linear. How are they different (linear Electron Geometry, linear Molecular Geometry, both linear EG and MG)?



- a. Explain the difference in polarity for HCN and CO_2 ?

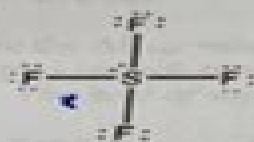
2. There are two ways to be bent. How are they different?

They're different by shape + angle of molecule



- a. What would be the formal charges, if SO_2 had a 1 single/1 double bond arrangement?
b. Are these polar or nonpolar molecules?
c. What shape is H_2O ?

3. Tetrahedral or not?



- a. Why are two molecules nonpolar and one molecule polar?

SF_4 is polar being that electrons are throwing off electronegativity.

4. What makes trigonal pyramidal and trigonal planar molecules different? *causing change in shape*



- a. Why are two of the molecules polar and one molecule nonpolar?

Pogil Intermolecular Forces And Boiling Points Answers

Richard Samuel Moog



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Process Oriented Guided Inquiry Learning (POGIL) Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes

Chemists' Guide to Effective Teaching Norbert J. Pienta, Melanie M. Cooper, Thomas J. Greenbowe, 2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Pogil Intermolecular Forces And Boiling Points Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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