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## POGIL-The Mole Concept (How can atoms be counted?)

### Why?

Consider the following equation for a chemical reaction:  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

This can be interpreted as two molecules of hydrogen and one molecule of oxygen combining to form two water molecules. But how often do chemists limit their reactions to one or two molecules? Usually a reaction is done with an unimaginable number of molecules. How then do chemists know they have the right mix? The molecules need to be quickly counted! How do we count molecules? The answer is the unit called the **mole**. Weighing is often easier than counting!

### Model 1: The Balanced Equation

How can the coefficients in a chemical equation be interpreted?

Consider the reaction:



### Key Questions

1. What are the coefficients for each of the following substances in the reaction?

### Read This!

Coefficients tell you the relative amounts of products and reactants. A balanced chemical equation can be interpreted in terms of different quantities, including:

- Numbers of particles (atoms, molecules, or formula units)

#### PARTICLES

Atoms



(if you have a single element)

- Mass
- Volume
- **Moles**

Molecules



(if you have a covalent compound)

Formula Units



(if you have an ionic compound)



# Pogil Introduction To Mole Concept

**Alan Keith Griffiths**



## **Pogil Introduction To Mole Concept:**

**Chemistry Education in the ICT Age** Minu Gupta Bhowon, Sabina Jhaumeer-Laulloo, Henri Li Kam Wah, Ponnadurai Ramasami, 2009-07-21 th th The 20 International Conference on Chemical Education 20 ICCE which had rd th Chemistry in the ICT Age as the theme was held from 3 to 8 August 2008 at Le Meridien Hotel Pointe aux Piments in Mauritius With more than 200 participants from 40 countries the conference featured 140 oral and 50 poster presentations th Participants of the 20 ICCE were invited to submit full papers and the latter were subjected to peer review The selected accepted papers are collected in this book of proceedings This book of proceedings encloses 39 presentations covering topics ranging from fundamental to applied chemistry such as Arts and Chemistry Education Biochemistry and Biotechnology Chemical Education for Development Chemistry at Secondary Level Chemistry at Tertiary Level Chemistry Teacher Education Chemistry and Society Chemistry Olympiad Context Oriented Chemistry ICT and Chemistry Education Green Chemistry Micro Scale Chemistry Modern Technologies in Chemistry Education Network for Chemistry and Chemical Engineering Education Public Understanding of Chemistry Research in Chemistry Education and Science Education at Elementary Level We would like to thank those who submitted the full papers and the reviewers for their timely help in assessing the papers for publication th We would also like to pay a special tribute to all the sponsors of the 20 ICCE and in particular the Tertiary Education Commission <http://tec.intnet.mu> and the Organisation for the Prohibition of Chemical Weapons <http://www.opcw.org> for kindly agreeing to fund the publication of these proceedings **The Mole Concept** Paul Gilchrist, 1982\* The Mole Concept, 1983 SUMMARY A series of 6 programmes explaining the reasoning behind the development and acceptance of the mole concept The Mole Concept Alan Keith Griffiths, 1979

## **Pogil Introduction To Mole Concept** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the ability of words has become more evident than ever. They have the capability to inspire, provoke, and ignite change. Such could be the essence of the book **Pogil Introduction To Mole Concept**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall impact on readers.

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