



The three mole ratios derived from the balanced equation above are:

$$\frac{1 \text{ mol N}_2}{3 \text{ mol H}_2}$$

$$\frac{2 \text{ mol NH}_3}{1 \text{ mol N}_2}$$

$$\frac{3 \text{ mol H}_2}{2 \text{ mol NH}_3}$$

Pogil Mole Ratios N₂ G 3h₂ G 2nh₃ G

Lifelique



Pogil Mole Ratios N₂ + 3H₂ → 2NH₃ G:

Mole Ratios Lifeliqe, 2019 This lesson plan relates balanced chemical equations to everyday analogies defines stoichiometry and uses mole ratios to convert between amounts of substances in chemical reactions

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