



The three mole ratios derived from the balanced equation above are:

$$\frac{1 \text{ mol N}_2}{3 \text{ mol H}_2}$$

$$\frac{2 \text{ mol NH}_3}{1 \text{ mol N}_2}$$

$$\frac{3 \text{ mol H}_2}{2 \text{ mol NH}_3}$$

# Pogil Mole Ratios N<sub>2</sub> G 3h<sub>2</sub> G 2nh<sub>3</sub> G

**Lifelique**

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*Mole Ratios* Lifeliqe,2019 This lesson plan relates balanced chemical equations to everyday analogies defines stoichiometry and uses mole ratios to convert between amounts of substances in chemical reactions

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