

## PRE TEST

## Chapter 6 Test-General/Honors Chemistry

## Part 1: Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

- |       |              |                           |  |
|-------|--------------|---------------------------|--|
| C.2.2 | <u>  a  </u> | 1. cation                 | a. a positively charged ion  |
| C.2.2 | <u>  g  </u> | 2. electronegativity      | b. an element whose highest occupied <i>s</i> or <i>p</i> sublevels are partially filled                     |
| C.2.5 | <u>  j  </u> | 3. inner transition metal | c. the energy required to remove an electron from an atom  |
| C.2.6 | <u>  i  </u> | 4. alkali earth metal     | d. good conductor of heat and electric current   |
| C.2.6 | <u>  h  </u> | 5. noble gas              | e. any element whose highest occupied <i>s</i> sublevel and a nearby <i>d</i> sublevel contain electrons     |
| C.2.5 | <u>  b  </u> | 6. representative element | f. any nonmetal in group 7A  |
| C.2.2 | <u>  d  </u> | 7. metal                  | g. the tendency of an atom to attract electrons when it's in a compound (how much an atom wants an electron) |
| C.2.6 | <u>  f  </u> | 8. halogen                | h. an element in which the highest occupied <i>s</i> and <i>p</i> sublevels are completely filled            |
| C.2.5 | <u>  e  </u> | 9. transition metal       | i. an element in group 2A  |
| C.2.2 | <u>  c  </u> | 10. ionization energy     | j. an element whose highest <i>s</i> sublevel and a nearby <i>f</i> sublevel contain electrons               |

## Part 2: Multiple Choice

Choose the best answer and write its letter on the line.

- |       |              |                                      |                          |                              |
|-------|--------------|--------------------------------------|--------------------------|------------------------------|
| C.2.6 | <u>  b  </u> | 11. The alkali metals do not include | a. Li                    | c. Na                        |
|       |              |                                      | b. Ca                    | d. Rb                        |
| C.2.6 | <u>  c  </u> | 12. The element iodine, I, is a      | a. period 5 alkali metal | c. period 5 halogen          |
|       |              |                                      | b. period 4 halogen      | d. period 5 transition metal |

# World Of Chemistry Chapter Test Answers

**Marcia L. Gillette, Ira Blei, Wendy  
Gloffke, George Odian**



## **World Of Chemistry Chapter Test Answers:**

**World of Chemistry** Melvin D. Joesten, James L. Wood, 1999 The quality of human life is described in terms understanding to the non science major in this application oriented chemistry textbook This paperback offers a more concise less theoretical less mathematical version of the hardback version **WORLD OF CHEMISTRY** Second Edition *Student Topic Outline and Study Guide to Accompany World of Chemistry, Essentials, Joesten/Netterville/Wood* James O. Schreck, 1993 *Concepts of Earth Science & Chemistry Parent Lesson Plan* John Hudson Tiner, 2013-08-26 Concepts of Earth and Chemistry Course Description This is the suggested course sequence that allows one core area of science to be studied per semester You can change the sequence of the semesters per the needs or interests of your student materials for each semester are independent of one another to allow flexibility Semester 1 Earth Blending a creationism perspective of history with definitions of terms and identification of famous explorers scientists etc this book gives students an excellent initial knowledge of people and places encouraging them to continue their studies in depth Semester 2 Chemistry Chemistry is an amazing branch of science that affects us every day yet few people realize it or even give it much thought Without chemistry there would be nothing made of plastic there would be no rubber tires no tin cans no televisions no microwave ovens or something as simple as wax paper This book presents an exciting and intriguing tour through the realm of chemistry as each chapter unfolds with facts and stories about the discoveries of discoverers Find out why pure gold is not used for jewelry or coins Join Humphry Davy as he made many chemical discoveries and learn how they shortened his life See how people in the 1870s could jump over the top of the Washington Monument Exploring the World of Chemistry brings science to life and is a wonderful learning tool with many illustrations and biographical information [Survey of Science History & Concepts Parent Lesson Plan](#) ,2013-08-01 Survey of Science History Concepts Course Description Students will study four areas of science Scientific Mathematics Physics Biology and Chemistry Students will gain an appreciation for how each subject has affected our lives and for the people God revealed wisdom to as they sought to understand Creation Each content area is thoroughly explored giving students a good foundation in each discipline Semester 1 Math and Physics Numbers surround us Just try to make it through a day without using any It s impossible telephone numbers calendars volume settings shoe sizes speed limits weights street numbers microwave timers TV channels and the list goes on and on The many advancements and branches of mathematics were developed through the centuries as people encountered problems and relied upon math to solve them It s amazing how ten simple digits can be used in an endless number of ways to benefit man The development of these ten digits and their many uses is the fascinating story in Exploring the World of Mathematics Physics is a branch of science that many people consider to be too complicated to understand John Hudson Tiner puts this myth to rest as he explains the fascinating world of physics in a way that students can comprehend Did you know that a feather and a lump of lead will fall at the same rate in a vacuum Learn about the history of physics from Aristotle to Galileo to

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Introduction to Organic Chemistry William H. Brown, Thomas Poon, 2016-01-13 Introduction to Organic Chemistry 6th Edition provides an introduction to organic chemistry for students who require the fundamentals of organic chemistry as a requirement for their major It is most suited for a one semester organic chemistry course In an attempt to highlight the relevance of the material to students the authors place a strong emphasis on showing the interrelationship between organic chemistry and other areas of science particularly the biological and health sciences The text illustrates the use of organic chemistry as a tool in these sciences it also stresses the organic compounds both natural and synthetic that surround us in everyday life in pharmaceuticals plastics fibers agrochemicals surface coatings toiletry preparations and cosmetics food additives adhesives and elastomers This text is an unbound three hole punched version Access to WileyPLUS sold separately

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**Gaither's Dictionary of Scientific Quotations** Carl C. Gaither, Alma E. Cavazos-Gaither, 2008-01-08 Scientists and other keen observers of the natural world sometimes make or write a statement pertaining to scientific activity that is destined to live on beyond the brief period of time for which it was intended. This book serves as a collection of these statements from great philosophers and thought influencers of science past and present. It allows the reader quickly to find relevant quotations or citations. Organized thematically and indexed alphabetically by author, this work makes readily available an unprecedented collection of approximately 18 000 quotations related to a broad range of scientific topics.

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**Brown's Introduction to Organic Chemistry** William H. Brown, Thomas Poon, 2017-06-28 Introduction to Organic Chemistry 6th Global Edition provides an introduction to organic chemistry for students who require the fundamentals of organic chemistry as a requirement for their major. It is most suited for a one semester organic chemistry course. In an attempt to highlight the relevance of the material to students, the authors place a strong emphasis on showing the interrelationship between organic chemistry and other areas of science, particularly the biological and health sciences. The text illustrates the use of organic chemistry as a tool in these sciences; it also stresses the organic compounds, both natural and synthetic, that surround us in everyday life: in pharmaceuticals, plastics, fibers, agrochemicals, surface coatings, toiletry preparations, and cosmetics, food additives, adhesives, and elastomers.

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