STOICHIOMETRY: MIXED PROBLEMS

Name _____

1. $N_2 + 3H_2 \rightarrow 2NH_3$

What volume of NH, at STP is produced if 25.0 g of $\rm N_2$ is reacted with an excess of $\rm H_2?$

2. 2KCIO₃ → 2KCI + 3O₉

If 5.0 g of KCIO, is decomposed, what volume of O_2 is produced at STP?

- 3. How many grams of KCI are produced in Problem 2?
- 4. Zn + 2HCl → ZnCl₂ + H₂

What volume of hydrogen at STP is produced when 2.5 g of zinc react with an excess of hydrochloric acid?

5. H,SO₄ + 2NaOH → H₂O + Na₃SO₄

How many molecules of water are produced if 2.0 g of sodium sulfate are produced in the above reaction?

6. 2AICI, → 2AI + 3CI,

If 10.0 g of aluminum chloride are decomposed, how many molecules of Cl₂ are produced?

Temperature And Its Measurement Chemistry If8766

G Thomas

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