

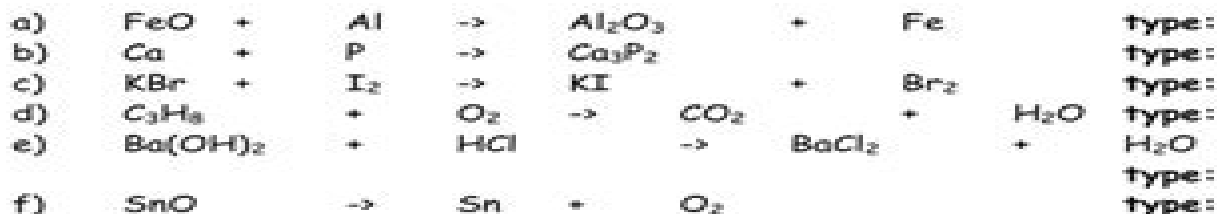
Grade 10 Chemistry Review

- Use the periodic table to answer the following questions concerning calcium.
 - In what period is it located?
 - In what group is it located?
 - What is its atomic number?
 - What is its atomic mass?
 - In what state would it be found at room temperature?
- Explain why O forms a -2 charged ion. Provide a written explanation and Bohr-Rutherford diagrams for both the atom and the ion.
- Write formulas for each of the following compounds.

sulfur pentafluoride
calcium oxide
nitrogen gas
iron (III) sulfide

sodium sulfate
tin(II) bromide
calcium phosphate
ammonia

- Name the type of reaction and balance each equation.



- Write word equations for each of the reactions in #4. C_3H_8 is propane.
- Describe a standard test for each of the following substances:
 - oxygen gas
 - hydrogen gas
 - carbon dioxide gas
 - water vapour

Tenth Grade Chemistry

E Durkheim



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