

Name: _____

Date: _____

Student Exploration: Temperature and Particle Motion

Vocabulary: absolute zero, Kelvin scale, kinetic energy, Maxwell-Boltzmann distribution, molar mass, molecule, temperature, universal gas constant

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Why is hot air hot? _____

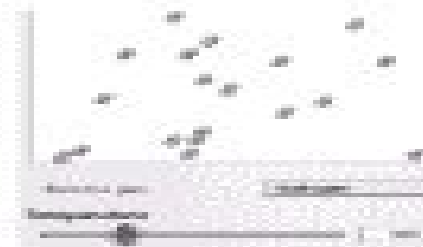
2. Why is cold air cold? _____

3. Air consists of tiny particles called **molecules**. How do you think the molecules move in hot and in cold air? _____

Gizmo Warm-up

The Temperature and Particle Motion Gizmo™ illustrates how the molecules of gas move at different temperatures. In this Gizmo, temperature is measured on the **Kelvin scale**, which measures temperature from **absolute zero**, the coldest possible temperature ($-273.15\text{ }^{\circ}\text{C}$). Each unit on the Kelvin scale is equivalent to $1\text{ }^{\circ}\text{C}$: $273.15\text{ K} = 0\text{ }^{\circ}\text{C}$, and $373.15\text{ K} = 100\text{ }^{\circ}\text{C}$.

Check that the selected gas is **Hydrogen** and the **Temperature** is **300 K**.



1. Describe the motion of the hydrogen molecules: _____

2. Are all of the molecules moving at the same speed? _____

Temperature And Particle Motion Gizmo Answers

Terry C. Jones

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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