

Assignment

The Periodic Law

Section Quiz: History of the Periodic Table

As Morgan's assistant, when presented with the periodic table below, please read each statement, mark statement as being correct or not (question).

1. In developing his periodic table, Mendeleev's observations were
 - a. correct.
 - b. incorrect and significant.
 - c. lacking.
 - d. **incomplete.**
2. Mendeleev's periodic table distinguishes all elements in order of increasing atomic mass because he grouped together elements with
 - a. **similar properties.**
 - b. similar structures.
 - c. masses.
 - d. charges.
3. Mendeleev predicted that the gaps in his periodic table represented
 - a. ions.
 - b. unknown elements.
 - c. possible isotopes.
 - d. **unidentified elements.**
4. The periodic table is a hybrid of periodic table based on increasing
 - a. atomic mass.
 - b. **atomic number.**
 - c. electronegativity.
 - d. **charge.**
5. The periodic law describes the physical and chemical properties of elements and predicts descriptions of their atoms.
 - a. **Correct.**
 - b. incorrect.
 - c. false.
 - d. incorrect.

Periodic Law Section Quiz Answers

Wolfgang Guggemos



Periodic Law Section Quiz Answers:

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Periodic Law Section Quiz Answers :

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