Use the below periodic table to answer the following questions:

| А | 1 | | Periodic Table of the Elements | | | | | | | | | | | | | D | |
|-----|---|-----|--------------------------------|--|--|--|---|--|-----|--|-----|---|---|---|---|---|---|
| | В | 1 | | | | | | | 140 | | 300 | С | - | + | 1 | | |
| | E | | | | | | | | F | | | - | - | + | | | |
| - 1 | G | | | | | | | | | | | | + | | + | | |
| | | +0 | | | | | | | | | | - | - | + | - | + | 1 |
| | | -20 | | | | | | | | | | | | | | - | |
| | | | | | | | - | | | | | | | | | | |

- a) Which element is located in group 2A, period 5?
- b) What is the period and the group of element (F)?
- c) Which element has the highest Electronegativity?
- d) Arrange (B, E, G) atoms in order of decreasing atomic radius.
- e) Arrange (B+2, C+3, D-1) ions in order of decreasing atomic radius.
- f) Write the electron configration of element (D) and its anion (D').
- g) Write the reaction of element (A) with water.

Periodic Table Test Questions And Answers

Neil D. Jespersen, Pamela Kerrigan

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