

Chemical Periodicity

Name Mitchell Wade Section Number 015

OBJECTIVE: To demonstrate the periodic law.

CONCEPTS: Periodic law, Periodic trends

INTRODUCTION: In the 1860's about 60 elements were known. From their physical properties, the elements were classified as metals, nonmetals, or metalloids. Metals have low electrical resistance (high conductivity), have a metallic luster, and are malleable. Non-metals have high resistance, low luster, and are brittle. Metalloids have properties between those of non-metals and metals. Chemists of the time further classified the elements into groups. The elements in a particular group have similar chemical and/or physical properties. This classification led to the discovery of the periodic table and the periodic law. In the periodic table, elements are assigned to a period and a group—groups are the columns and the periods are the rows of the table. The periodic law states that the properties of the chemical elements vary with the atomic number in a systematic way. We now understand that the properties depend on the electronic structures of the atoms.

PROCEDURE:

1. Find the station labeled "Halogens: states of matter." Observe the states of matter of the three samples. How does the state of matter change as you go down the column of the periodic table?

Starts as gas and goes to solid

The melting points of the halogen elements that you observed are listed below. Based on the trend that you see, estimate the melting point of Fluorine.

Element	Melting Point ($^{\circ}\text{C}$)
F	-200
Cl	-101.5
Br	-7.35
I	113.75

2. Find the station labeled "Metallic Solutions." Fill out the chart below. For the unlabeled mystery elements, you will have to guess which type of metal they are (transition, alkali, or alkaline earth) based on the trends you see.

Solution Number	Element	Color of Water Solution	Type (transition metal, alkali metal, alkaline earth metal)
1	Co	Red	Transition metal
2	Ba	Clear	Alkaline earth metal
3	K	clear	Alkali metal
4	Ni	green	Transition metal
5	Na	clear	Alkali metal
6	Cu	blue	Transition metal
7	Mg	clear	Alkaline earth metal

Periodicity Lab Report

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Table of Contents Periodicity Lab Report

1. Understanding the eBook Periodicity Lab Report
 - The Rise of Digital Reading Periodicity Lab Report
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodicity Lab Report
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodicity Lab Report
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodicity Lab Report
 - Personalized Recommendations
 - Periodicity Lab Report User Reviews and Ratings
 - Periodicity Lab Report and Bestseller Lists
5. Accessing Periodicity Lab Report Free and Paid eBooks
 - Periodicity Lab Report Public Domain eBooks
 - Periodicity Lab Report eBook Subscription Services
 - Periodicity Lab Report Budget-Friendly Options

6. Navigating Periodicity Lab Report eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodicity Lab Report Compatibility with Devices
 - Periodicity Lab Report Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodicity Lab Report
 - Highlighting and Note-Taking Periodicity Lab Report
 - Interactive Elements Periodicity Lab Report
8. Staying Engaged with Periodicity Lab Report
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodicity Lab Report
9. Balancing eBooks and Physical Books Periodicity Lab Report
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodicity Lab Report
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodicity Lab Report
 - Setting Reading Goals Periodicity Lab Report
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodicity Lab Report
 - Fact-Checking eBook Content of Periodicity Lab Report
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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