

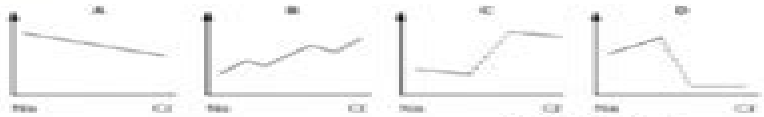
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## I. Periodicity

2002 -2019

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1. Which species represented by the following formulae has the largest radius?
- A.  $P^{3-}$       B.  $Cl$       C.  $Ar$       D.  $K^+$  [2002 M/1 (1.2)]
2. Which of the following oxides is *unlikely* to dissolve in aqueous sodium hydroxide?
- A.  $Al_2O_3$       B.  $MgO$       C.  $SO_2$       D.  $SiO_2$  [2002 M/1 (1.2)]
3. An element of the third period (Na to Cl) is heated in chlorine. The product is purified and then added to water. The resulting solution is found to be neutral.
- What is the element?
- A. sodium  
B. aluminium  
C. silicon  
D. phosphorus [2002 M/1 (1.4)]
4. The chloride of element Q is hydrolysed by water to form an acidic solution and its oxide reacts with acid to form a salt.
- What could be the element Q?
- A. magnesium  
B. aluminium  
C. silicon  
D. phosphorus [2002 Q/18 (1.3)]

5. Which diagram represents the change in ionic radius of the elements across the third period (Na to Cl)?
- 
- [2002 Q/18 (1.4)]
6. Which isotope of an element in the third period of the Periodic Table contains the same number of neutrons as  $^{32}_{16}S$ ?
- A.  $^{23}_{11}Na$   
B.  $^{24}_{12}Mg$   
C.  $^{28}_{14}Si$   
D.  $^{31}_{15}P$  [2002 M/1 (1)]
7. The successive ionisation energies, in  $\text{kJ mol}^{-1}$ , of an element X are given below.
- | $x/\text{e}$                 | 1st  | 2nd  | 3rd  | 4th  | 5th  | 6th    | 7th |
|------------------------------|------|------|------|------|------|--------|-----|
| Energy/ $\text{kJ mol}^{-1}$ | 1400 | 2800 | 4600 | 5400 | 7000 | 13,100 |     |
- What is X?
- A.  $_{11}\text{Na}$       B.  $_{12}\text{Mg}$       C.  $_{13}\text{Al}$       D.  $_{14}\text{Si}$  [2002 M/1 (1.4)]

# Periodicity Multiple Choice Questions

**Wenbin Ji**



## **Periodicity Multiple Choice Questions:**

This book delves into Periodicity Multiple Choice Questions. Periodicity Multiple Choice Questions is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Periodicity Multiple Choice Questions, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Periodicity Multiple Choice Questions
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- Chapter 4: Periodicity Multiple Choice Questions in Specific Contexts
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6. In chapter 5, the author will draw a conclusion about Periodicity Multiple Choice Questions. This chapter will summarize the key points that have been discussed throughout the book.

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