

# Unit: Chemical Bonding Answer Key "Covalent Bonding" - WS #3

Directions: Please answer each question using your notes from today's class. 

©

 What is the difference between a molecular formula, structural formula and an electron dot formula? Give an example of each.

A molecular formula gives the chemical make-up of one molecule. It does not show atom-to-atom connectivity. Example:  $C_2H_2$ 

A structural formula uses lines instead of dots to show shared pairs of electrons.

Example: H-H

An electron dot formula uses dots instead of lines to show shared pairs of electrons. Example: H:H

2. What is the difference between single, double, and triple covalent bonds? How many shared electrons are in each?

The difference between the types of bonds is: a single bond is two shared electrons, a double is four shared electrons, and a triple is six shared electrons.

Draw the electron dot and structural formulas for the following molecules.

Substance	Rough copy (work!)	Electron dot	Structural
a) Cl <sub>2</sub>		Note: The electron dot structures have dots in between each atom instead of lines.	:çı—çı:
b) H <sub>2</sub> S			:\$-н Н
c) Brl			ij <u></u> —₿ŗ:

# **Review Answers For Covalent Bonding**

John Moore, Richard H. Langley

#### **Review Answers For Covalent Bonding:**

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Unveiling the Magic of Words: A Review of "Review Answers For Covalent Bonding"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "Review Answers For Covalent Bonding," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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