

## → CHEMICAL EQUILIBRIUM

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**Q.1:** Justify that chemical equilibrium is dynamic in nature?

**Ans:** In reversible chemical reactions the molecules of reactions collide and convert into products. At the same time the molecules of the products are converted into reactants. When two opposing forces maintain the equal, rates then equilibrium is there and that is dynamic equilibrium in nature.

**Q.2:** When does four types of chemical equilibrium constants for a reaction become equal?

**Ans:** When we have such a reversible reaction, in which the total number of moles of reactants and products are equal then  $K_p = K_c = K_n$ . In other words  $\Delta n = 0$ , where  $\Delta n$  is the difference of number of moles of reactants and products.

**Q.3:** The change of temperature disturbs the equilibrium position and the equilibrium constant of reaction? Justify?

**Ans:** All the reversible reactions are disturbed by changing their equilibrium position and equilibrium constant by disturbing the temperature. Actually change of temperature changes the energy contents of reactants and products.

**Q.4:** Why is solubility of glucose in water is increased by increasing the temperature?

**Ans:** The solubility of glucose in water is an endothermic process. Increase of temperature pushed the system to that side where heat is absorbed, and that is the side where solubility increases.

**Q.5:** What is the nature of solution having  $\text{pH} = 7.5$ ?

**Ans:** The solution with  $\text{pH} = 7$  is neutral and having values more than 7 have greater concentration of  $\text{OH}^-$  and less concentration of  $\text{H}^+$ , so the solution is basic.

**Q.6:** What is buffer capacity?

**Ans:** Buffer capacity is a ability of a buffer to resist the change of  $\text{pH}$ , when a few drops of an acid and base is added from outside. The best buffer is obtained. (i) when  $\text{pH}$  of the buffer is equal to  $\text{pK}_a$  of the acid and the concentration of the salt and acid are equal. (ii) The  $\text{OH}^-$  of the buffer is equal to  $\text{pK}_b$  or the base.

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*The Origin and Nature of Life on Earth* Eric Smith, Harold J. Morowitz, 2016-03-31 Uniting the conceptual foundations of the physical sciences and biology this groundbreaking multidisciplinary book explores the origin of life as a planetary process Combining geology geochemistry biochemistry microbiology evolution and statistical physics to create an inclusive picture of the living state the authors develop the argument that the emergence of life was a necessary cascade of non equilibrium phase transitions that opened new channels for chemical energy flow on Earth This full colour and logically structured book introduces the main areas of significance and provides a well ordered and accessible introduction to multiple literatures outside the confines of disciplinary specializations as well as including an extensive bibliography to provide context and further reading For researchers professionals entering the field or specialists looking for a coherent overview this text brings together diverse perspectives to form a unified picture of the origin of life and the ongoing organization of the biosphere

*Science Reporter*, 1989

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