

10.1

THE NATURE OF GASES

SECTION REVIEW

Objectives

- Describe the motion of gas particles according to the kinetic theory
- Interpret gas pressures in terms of kinetic theory

Key Terms

- kinetic energy
- kinetic theory
- gas pressure
- vacuum
- standard atmospheric pressure
- atmospheric pressure
- barometer

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are highlighted in this section. Write the blank over the completed word or phrase, if one is possible, or a synonym.

- The kinetic theory describes the _____ of particles in matter and the forces of attraction between them. The theory assumes that the volume occupied by a gas is mostly _____, that the particles of gas are _____, move _____ of each other, and are in constant _____ motion. Furthermore, the _____ between particles are elastic so that the total _____ remains constant. The pressure of a gas results from the countless collisions of the gas particles with an object. Barometers are used to measure _____ pressure. Standard conditions are defined as a temperature of _____ and a pressure of _____.
- _____
 - _____
 - _____
 - _____
 - _____
 - _____
 - _____
 - _____
 - _____
 - _____

Part B True-False

Classify each of these statements as true or false (true, *AT*; sometimes true, *ST*; or never true, *NT*).

- Atmospheric pressure is 760 mm Hg.
- The SI unit of pressure is the pascal.
- Atmospheric pressure increases as you climb a mountain because the air layer that compresses the air gets thicker with height.

Review Gases Section 4 Answers

Yi-Tong Ma

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Review Gases Section 4 Answers:

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