

# Chapter 3 Review Questions

Question: 2, 4, 12 (d), 12 (e), 16 (d), 16 (e), 18 (d), 18 (e), 26, 34 (a), 34 (c), 38, 40

2. What is the total mass of hydrogen in each of the molecules? *Atomic mass (H) = 1.008 g/mol*
- $\text{CH}_4$
  - $\text{CHCl}_3$
  - $\text{C}_{12}\text{H}_{10}\text{O}_6$
  - $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$

a)  $H = 4$

Total mass (H) =  $4 \times 1.008 \text{ g/mol}$

Total mass (H) = 4.032 amu

a)  $H = 12$

Total mass (H) =  $12 \times 1.008 \text{ g/mol}$

Total mass (H) = 12.096 amu

b)  $H = 1$

Total mass (H) =  $1 \times 1.008 \text{ g/mol}$

Total mass (H) = 1.008 amu

c)  $H = 10$

Total mass (H) =  $10 \times 1.008 \text{ g/mol}$

Total mass (H) = 10.08

4. Determine the molecular mass of the following compounds:

(a)



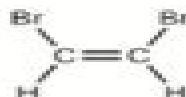
Molar mass =  $(2 \times 35.45) + (1 \times 12.01) + (1 \times 15.9) = 98.81 \text{ g/mol}$

(b)



Molar mass =  $(2 \times 1.008) + (2 \times 12.01) = 26.038 \text{ g/mol}$

(c)



Molar mass =  $(2 \times 1.008) + (2 \times 79.904) + (2 \times 12.01) = 165.84 \text{ g/mol}$

(d)



Molar mass =  $(2 \times 16.00) + (4 \times 15.999) + (2 \times 1.008) = 98.078 \text{ g/mol}$

# Review Questions For Hewitt Chapte 35

**Albert A Gayle**



## **Review Questions For Hewitt Chapte 35:**

## Reviewing **Review Questions For Hewitt Chapte 35**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Review Questions For Hewitt Chapte 35**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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