Revision - Physical Chemistry -3a) Energetics
Define the following reactions and give an example of each  Ensothermic <u>releasey hearf to the environment eng.</u> Burning,  Exothermic Takes in heart eng. Meldra ice:
Explain what experiment you can do to determine how much energy is given out during neutralization  Put Prichard in a polyshyrene up a menore componente  at the start + end asing a thermometer
Praw a labelled energy level diagram to show an exothermic reaction.  Rachsuh  products  products
Using the specific heat capacities water: 4200 J/kg *C
<ol> <li>Excess iron powder was added to 100.0 cm² of 0.200 mol dm² copper(II) sulfate solution in a calorimeter. The reaction equation was as follows. Fe (s) + CuSO.(aq) → FeSO.(aq) + Cu (s). The maximum temperature rise was 7.5 °C. Determine the heat energy change of the reaction, in kJ.</li> <li>□ = fn x C x fx</li> </ol>

 1.023 g of propen-1-ol (M = 60.11 g mol\*) was burned in a spirit burner and used to heat 200 g of water in a copper calorimeter. The temperature of the water rose by 30 °C. Calculate the heat energy change for the combustion of propen-1-ol using this data.

 A chemist uses a Bunsen burner to heat some ethanol in a beaker. If the energy input is 2.5kJ and the temperature change is 21°C to 32°C, find the mass of ethanol that was heated in kg.

3. A spinning bicycle wheel needs to be brought to a stop by a rubber brake block. Assuming the wheel has 500u of kinetic energy that is all absorbed by the brake and that the mass of the block is 25g, how much will the temperature of the block rise by in "C?"

# **Review Reaction Rates And Equilibrium Answers**

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#### **Review Reaction Rates And Equilibrium Answers:**

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Whispering the Strategies of Language: An Mental Quest through Review Reaction Rates And Equilibrium Answers

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