

Name:

Period:

### Chemical Bonding Worksheet

Ionic Bond  
Covalent Bond  
Metallic Bond

between a Metal and Non-Metal  
between a Non-Metal and Non-Metal  
between a Metal and Metal

(M + NM)  
(NM + NM)  
(M + M)

Determine if the elements in the following compounds are metals or non-metals. Describe the type of bonding that occurs in the compound.

Compound	Element 1 (metal or non-metal?)	Element 2 (metal or non-metal?)	Bond Type	Formula Mass (g/mol)
NO <sub>2</sub>	N = non-metal	O = non-metal	covalent	
NaCl				
SO <sub>2</sub>				
PO <sub>4</sub> <sup>3-</sup>				
MgBr <sub>2</sub>				
CaO				
H <sub>2</sub> O				
K <sub>2</sub> O				
Cu-Zn alloy				
O <sub>2</sub>				
CuCl <sub>2</sub>				
NO <sub>2</sub> <sup>-</sup>				
TiO <sub>2</sub>				
HF				
Rb <sub>2</sub> S				
Au-Ag mixture				
Fe <sub>2</sub> O <sub>3</sub>				
C <sub>6</sub> H <sub>12</sub> O <sub>22</sub>				

# Review Sheet Chemical Bonding Answers

**Kevin T. Patton, Gary A.  
Thibodeau, Linda Swisher**



## Review Sheet Chemical Bonding Answers:

**Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual** Leo J. Malone, Theodore O. Dolter, 2012-01-03

The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

**Student Study Guide** Peter L. Kresan, Reed

Mencke, Frank Press, Raymond Siever, 2003-09-25. This reconceptualization of the text *Understanding Earth* reflects the fundamental changes in the field of physical geology over the past several years.

*Organic Chemistry Study Guide and Solutions Manual* Marc Loudon, Jim Parise, Lisa Bonner, 2024-08-09. The Study Guide and Solutions Manual offers several study tools for students. Study Guide Links are brief supplements to text material and provide hints for study shortcuts, approaches to problem solving, or additional explanations of topics that typically cause difficulty. Further Explorations are short discussions that provide additional depth on content in the text. Reaction Reviews present the reactions in the chapter in detailed outline form. This summary consolidates the reactions of the chapter in one place and presents not only the reactions themselves but also the essential features of their mechanisms and stereochemistry for cases in which these issues are discussed in the text. Solutions are provided for all of the over 1600 problems in the text. Many include worked-out examples in which not only the answer is given but also the detailed approach to the solution is discussed.

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