

Names \_\_\_\_\_, Per. \_\_\_\_\_, Date \_\_\_\_\_

**Assignment: Thermochemistry Free Response Review**

1. Define thermochemistry and explain its importance in everyday life.
2. Explain the difference between endothermic and exothermic reactions. Give an example of each.
3. What is enthalpy? How does it relate to chemical reactions?
4. Define and differentiate between the system and the surroundings in the context of thermochemistry.
5. Explain what a calorimeter is and how it is used in thermochemistry.
6. What is Hess's Law? How is it applied in thermochemistry?
7. Calculate the heat released by 200g of water cooling from 75°C to 25°C. (Specific heat capacity of water is 4.18 J/g°C)
8. What is a state function, and why is enthalpy considered one?
9. Describe the first law of thermodynamics and its implications in chemical reactions.

# Response Questions For Thermochemistry High Schol

**Yijin Wang**



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