In the list below, circle the properties of acids and box the properties of bases.

Sour taste Feel slippery Contains hydroxide ions

Turns litmus blue Reacts with metals to form H Turns litmus red

Bitter taste Contains hydrogen ions Corrosive

Complete the chart below:

| Acid-Base Definitions |                               |                               |
|-----------------------|-------------------------------|-------------------------------|
| Type                  | Acid Definition               | Base Definition               |
| Arrhenius             | Donates an H* ion in solution | Donates an OH ion in solution |
| Bronsted-Lowry        | Donates an H* ion             | Accepts an H* ion             |
| Lewis                 | Accepts an electron pair      | Donates an electron pair      |

- 3. What term(s) describe a substance that can act as an acid or a base? Amphoteric or amphiprotic
- 4. Neutralization reactions always produce a SALT and WATER
- If the [H<sup>\*</sup>] is greater than the [OH<sup>\*</sup>], the solution is acidic basic neutral
- 6. If the [OH] is greater than the [H], the solution is acidic basic neutral
- 7. If the [H'] is equal to the [OH'], the solution is acidic basic neutral
- 8. If the pH is 7, the solution is acidic basic neutral
- If the pH is greater than 7, the solution is acidic basic neutral
- 10. If the pH is less than 7, the solution is acidic basic neutral
- 11. How many moles of lithium hydroxide are required to neutralize 19.2 grams of chloric acid?

$$LiOH + HClO_3 \rightarrow LiCl + H_2O$$

(19.2 g HClO<sub>2</sub>) x (1 mol HClO<sub>2</sub>/84 g HClO<sub>2</sub>) x (1 mol LiOH/1 mol HClO<sub>2</sub>) = 0.228 mol LiOH

12. How many grams of calcium hydroxide are required to neutralize 132.4 grams of nitric acid?

$$Ca(OH)_2 + 2 HNO_3 \rightarrow Ca(NO_3)_2 + 2 H_2O$$

(132.4 g HNO<sub>3</sub>) x (1 mol HNO<sub>3</sub>/63 g HNO<sub>3</sub>) x (1 mol Ca(OH)<sub>2</sub>/2 mol HNO<sub>3</sub>) x (74 g Ca(OH)<sub>2</sub>/1 mol Ca(OH)<sub>2</sub>) = 77.76 g Ca(OH)<sub>2</sub>

# **Review Acids And Bases Modern Chemistry Answers**

**M** Carnoy

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