### **Types of Chemical Reaction Worksheet**

Balance the reactions 1 to 6 and indicate which type of chemical reaction (synthesis, decomposition, single-displacement, double-displacement or combustion) is being represented:

<ol> <li>2 NaBr + Ca(OH)<sub>1</sub> → CaBr<sub>2</sub> + 2 NaOH</li> <li>Reaction Type : double displacement</li> </ol>	
	4.0

Indicate which type of chemical reaction (synthesis, decomposition, single-displacement, doubledisplacement or combustion) is being represented in 7 to 18.

# Review Activity Categories Of Chemical Reactions Answers

Morris Hein, Susan Arena, Cary Willard

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