

Molecular Composition of Gases

PROBLEM 1: PART 1 (25)

PROBLEM 1: Answer the following questions in the space provided.

1. _____ The average speed of a gas molecule is most directly related to the _____.
 - (a) volume of the container
 - (b) pressure within the gas
 - (c) temperature of the gas
 - (d) number of molecules in the gas
2. _____ Which value below best represents the average molecular speed of a diatomic gas given at the same temperature?
 - (a) The value associated with nitrogen gas is directly proportional.
 - (b) The value associated with nitrogen gas directly proportional.
 - (c) The value associated with oxygen is directly proportional.
 - (d) The value associated with oxygen is directly proportional.
3. _____ Which combination of conditions corresponds to the following reaction?

$$\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$$

 - a. _____ N_2 is present. The volume of H_2 is constant. The volume of NH_3 is constant.
 - b. _____ N_2 is present. The volume of H_2 is constant. The volume of NH_3 is constant.
4. _____ Which of the following is not a gas at room temperature?
 - a. _____ Methane
 - b. _____ Nitrogen
 - c. _____ Oxygen
 - d. _____ Carbon dioxide

PROBLEM 2: Answer the following questions in the space provided.

1. _____ Which of the following is not a gas at room temperature?
 - (a) Methane
 - (b) Nitrogen
 - (c) Oxygen
 - (d) Carbon dioxide

Review Molecular Composition Of Gases Short Answer

Emilie Sanchez



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