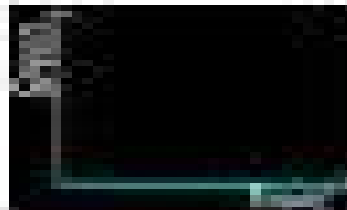


2nd Year Honours BSc in Applied Science, University of Dundee
 CH101

This assignment will require you to use your knowledge of the following topics to answer the questions. You should refer to your lecture notes and any relevant textbooks. You should also refer to the relevant chapters in the following books: Principles of Chemistry by G. L. Fisher and Principles of Chemistry by R. A. Scott.

Questions

- 1) Define the following terms: **enthalpy** **entropy** **free energy** **equilibrium constant**
- 2) Define the difference between enthalpy and entropy? **Enthalpy is a measure of heat energy, entropy is a measure of disorder.**
- 3) Define the difference between enthalpy and free energy? **Enthalpy is a measure of heat energy, free energy is a measure of the energy available to do work.**
- 4) Explain, using an example, why a reaction can be spontaneous even if it is endothermic. **Entropy is a measure of disorder, and a reaction can be spontaneous if the increase in entropy is greater than the increase in enthalpy.**
- 5) Define the following terms: **enthalpy of formation** **enthalpy of combustion** **enthalpy of atomization** **enthalpy of hydration** **enthalpy of solution** **enthalpy of vaporization**
- 6) How does the sign of the enthalpy change of a reaction affect the sign of the entropy change? **Enthalpy change is a measure of heat energy, and entropy change is a measure of disorder. A reaction with a positive enthalpy change and a positive entropy change is more likely to be spontaneous than a reaction with a positive enthalpy change and a negative entropy change.**
- 7) How does the sign of the enthalpy change of a reaction affect the sign of the free energy change? **Free energy change is a measure of the energy available to do work, and it is affected by both enthalpy and entropy changes. A reaction with a positive enthalpy change and a positive entropy change is more likely to be spontaneous than a reaction with a positive enthalpy change and a negative entropy change.**
- 8) Define the terms: **enthalpy of formation** **enthalpy of combustion** **enthalpy of atomization** **enthalpy of hydration** **enthalpy of solution** **enthalpy of vaporization**



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Guillaume Favre



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Annu Kaushik,2023-05-25 Quick Review Series QRS for BSc Nursing Semester I and II 2nd edition is an extremely exam oriented book The content has been developed and arranged in a manner so the entire INC syllabus has been covered The subject content has been divided unit wise and according to the weightage of marks in each unit It is well illustrated with simple reproducible diagrams and flow charts To aid in quick learning before examinations memory aides and mnemonics have also been added The book will serve the requirements of BSc Nursing Semester I and II students to prepare for their examinations Content presented in well classified sections Compliant to the revised INC Curriculum All question types asked in the University exams are addressed Long essays Short essays and Short notes Clearly describes how to attempt a question in the exam Suitable number of tables and flow charts according to the need of the questions Reproducible diagrams in Anatomy and Physiology sections **Institute of Human Sciences of Boston College Review** ,1974 **Public Relations Review** Foundation for Public Relations Research and Education (U.S.),2002 *Popular Photography* ,1984-10 **Proceedings** American Society for Engineering Education. Conference,1990 *Trade Policy Review* ,2017 *A Guide to Educational Programs in Noncollegiate Organizations* ,1980 *Proceedings* American Society for Engineering Education,1990 **Mosby's Comprehensive Review of Critical Care** Donna A. Zschoche,1981 **College Credit Recommendations** ,1993 *The Best 296 Business Schools, 2013 Edition* Princeton Review,2012-10-09 Provides a detailed overview of the best business schools across North America including information on each school s academic program competitiveness financial aid admissions requirements and social scenes Original **Report on Chilean University Life** ,1981 *Technological Advancement Through Canada-U.S.-global Interchange* American Society for Engineering Education. Conference,1990 *The Best 294 Business Schools* Princeton Review (Firm),2011 Provides a detailed overview of the best business schools across North America including information on each school s academic program competitiveness financial aid admissions requirements and social scenes *New York Magazine* ,1991-08-12 New York magazine was born in 1968 after a run as an insert of the New York Herald Tribune and quickly made a place for itself as the trusted resource for readers across the country With award winning writing and photography covering everything from politics and food to theater and fashion the magazine s consistent mission has been to reflect back to its audience the energy and excitement of the city itself while celebrating New York as both a place and an idea *Cadence* Bob Rusch,1990 **Theses and Dissertations** Catholic University of America. Library,1970 **ERASMUS Newsletter** ,1989 *Indiana Pharmacist* ,1928

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